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(72) Inventors:  
• Harris, Laurence James  
Sandwich, Kent CT13 9NJ (GB)  
• Levett, Philip Charles  
Sandwich, Kent CT13 9NJ (GB)

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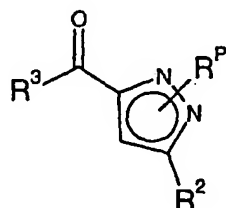
(74) Representative:  
Simpson, Alison Elizabeth Fraser  
Urquhart-Dykes & Lord, 30 Welbeck Street  
London W1G 8ER (GB)

(71) Applicants:  
• PFIZER INC.  
New York, N.Y. 10017 (US)  
Designated Contracting States:  
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• Pfizer Limited  
Sandwich, Kent CT13 9NJ (GB)  
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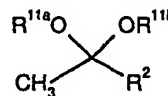
(54) Process for the preparation of pyrazoles

(57) There is provided a process for the production of pyrazole compounds of general formula (II):

wherein  $R^P$  represents H or  $R^1$ ; which process comprises the reaction of a compound of general formula (III),



II



III

wherein  $R^{1a}$  and  $R^{1b}$  independently represent  $C_1$ - $C_6$  alkyl,  $R^2$  is as defined herein before, with an acylating agent in the presence of a base, followed by *in situ* reaction with a hydrazine compound, wherein said process is optionally carried out in the presence of an activating agent.

## Description

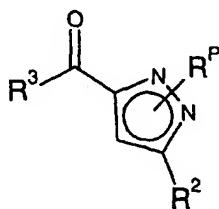
[0001] This invention relates to a novel process for the preparation of pyrazoles. In particular the present invention relates to a novel process for the preparation of pyrazole intermediates useful in the synthesis of 4-alkylpiperazinyl-sulfonylphenyl- and 4-alkylpiperazinylsulfonyl pyridinyldihydropyrazolo[4,3-d]pyrimidin-7-one derivatives which are potent and selective cGMP PDE<sub>5</sub> inhibitors.

[0002] A general synthetic route for the preparation of 4-alkylpiperazinylsulfonylphenyl-dihydropyrazolo[4,3-d]pyrimidin-7-one derivatives (as described in EP 812 845, EP 994 115 and WO98/49166, and for analogues thereof as described in WO 99/54333). This synthesis involves a coupling reaction between an intermediate pyrazole compound (I) (as illustrated in scheme 1) and a phenyl or pyridinyl derivative followed by cyclisation of the resulting coupled intermediate to provide pyrimidin-7-ones.

[0003] We have now found that pyrazole compounds of general formula (II) which are used to prepare the intermediate pyrazole compounds (I) may be made *via* a novel "one-pot" process, as described hereinafter and illustrated in Scheme 1, which process has advantages over the multi-step processes for the preparation of compounds of general formula (II) as described in EP 812 845, EP 994 115, WO 98/49166 and WO 99/54333.

[0004] A further synthetic route for the preparation of certain pyrazole compounds of general formula (II) is described in Martins, M. A. P.; Freitag, R.; Flores, A. F. C.; Zanatta, N. *Synthesis*, 1995, 1491 and Martins, M. A. P.; Flores, A. F. C.; Zanatta, N.; Bastos, G. P.; Bonacorso, H. G.; Siqueira, G. M. *Tetrahedron Lett.* 1999, 40, 4309. In this processes the pyrazole compounds are prepared in two steps.

[0005] According to a first aspect of the invention, there is provided a novel "one-pot" process for the production of pyrazole compounds of general formula (II):



II

wherein R<sup>P</sup> represents H or R<sup>1</sup>;

R<sup>1</sup> represents: H, C<sub>1</sub>-C<sub>6</sub> alkyl; C<sub>1</sub>-C<sub>6</sub> alkoxy; C<sub>3</sub>-C<sub>6</sub> cycloalkyl; C<sub>1</sub>-C<sub>6</sub> alkyl(C<sub>1</sub>-C<sub>6</sub> alkoxy), Het, C<sub>1</sub>-C<sub>6</sub> alkylHet, aryl or C<sub>1</sub>-C<sub>6</sub> alkylaryl, which latter eight groups are all optionally substituted (and/or, in the case of C<sub>1</sub>-C<sub>6</sub> alkyl, optionally terminated) by one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, C(O)NR<sup>4</sup>R<sup>5</sup>, C(O)R<sup>6</sup>, C(O)OR<sup>7</sup>, OR<sup>8</sup>, NR<sup>9a</sup>R<sup>9b</sup> and SO<sub>2</sub>NR<sup>10a</sup>R<sup>10b</sup>;

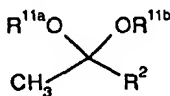
R<sup>2</sup> represents C<sub>1</sub>-C<sub>6</sub> alkyl; C<sub>1</sub>-C<sub>6</sub> alkoxy; C<sub>3</sub>-C<sub>6</sub> cycloalkyl; C<sub>1</sub>-C<sub>6</sub> alkyl(C<sub>1</sub>-C<sub>6</sub> alkoxy), Het, C<sub>1</sub>-C<sub>6</sub> alkylHet, aryl or C<sub>1</sub>-C<sub>6</sub> alkylaryl, which latter eight groups are all optionally substituted (and/or, in the case of C<sub>1</sub>-C<sub>6</sub> alkyl, optionally terminated) by one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, C(O)NR<sup>4</sup>R<sup>5</sup>, C(O)R<sup>6</sup>, C(O)OR<sup>7</sup>, OR<sup>8</sup>, NR<sup>9a</sup>R<sup>9b</sup> and SO<sub>2</sub>NR<sup>10a</sup>R<sup>10b</sup>;

R<sup>3</sup> represents: OH, C<sub>1</sub>-C<sub>6</sub> alkoxy or NR<sup>4</sup>R<sup>5</sup>;

R<sup>4</sup>, R<sup>5</sup>, R<sup>6</sup>, R<sup>7</sup>, R<sup>8</sup>, R<sup>10a</sup> and R<sup>10b</sup> independently represent H or C<sub>1</sub>-C<sub>6</sub> alkyl;

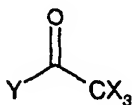
R<sup>9a</sup> and R<sup>9b</sup> either independently represent, H or C<sub>1</sub>-C<sub>6</sub> alkyl or, together with the nitrogen atom to which they are attached, represent azetidiny, pyrrolidinyl or piperidinyl,

which process comprises the reaction of a compound of general formula (III),



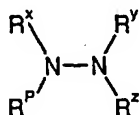
III

wherein R<sup>11a</sup> and R<sup>11b</sup> independently represent C<sub>1</sub>-C<sub>6</sub> alkyl, R<sup>2</sup> is as defined herein before, with an acylating agent of the formula (IV) in the presence of a base,



IV

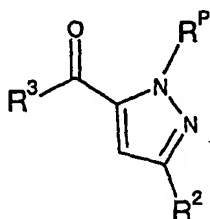
wherein X represents halogen independently selected from Cl, F or Br, and Y represents halogen or OR<sup>12</sup> wherein R<sup>12</sup> is: C<sub>1</sub>-C<sub>6</sub> alkyl; C(O)CX<sub>3</sub>; Het; or C<sub>1</sub>-C<sub>6</sub> alkyl(Het) wherein Het is pyridine or imidazole, followed by *in situ* reaction with a hydrazine compound of general formula (V)



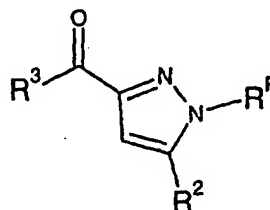
V

R<sup>p</sup> represents H or R<sup>1</sup> wherein R<sup>1</sup> is as defined hereinbefore, and R<sup>x</sup>, R<sup>y</sup> and R<sup>z</sup> are independently selected from H or electron donating groups (EDG) or electron withdrawing groups (EWG) wherein said electron withdrawing or donating groups are labile under the conditions of the reaction, wherein said process is optionally carried out in the presence of an activating agent, which process is referred to hereinafter as "the process of the invention". Advantageously, in the process according to the present invention the direct addition of the hydrazine compound (V) to the reaction vessel containing the reaction mixture of (III), (IV) and base results in production of compounds (II) with desirable purity and yield.

[0006] The compounds of general formula (II) can be represented by the formulae (IIA) and (IIB) as detailed hereinafter. The novel process according to the present invention includes the preparation of compounds of the formulae (IIA) and (IIB) as illustrated in Schemes 2 and 3.



IIA



IIB

[0007] The term "aryl", when used herein, includes six- to ten-membered carbocyclic aromatic groups, such as phenyl and naphthyl and the like.

[0008] Het groups may be fully saturated, partly unsaturated, wholly aromatic, partly aromatic and/or bicyclic in character. Het groups that may be mentioned include groups such as optionally substituted azetidiny, pyrrolidinyl, imidazolyl, indolyl, oxadiazolyl, thiadiazolyl, triazolyl, tetrazolyl, oxatriazolyl, thiatriazolyl, pyridazinyl, morpholinyl, pyrimidinyl, pyrazinyl, pyridyl, quinolinyl, isoquinolinyl, piperidinyl, pyrazolyl, imidazopyridinyl, piperazinyl, thienyl and furanyl.

[0009] The point of attachment of any Het group may be via any atom in the ring system including (where appropriate) a heteroatom. Het groups may also be present in the *N*- or *S*-oxidised form.

[0010] The term " $C_1$ - $C_6$  alkyl" (which includes the  $C_1$ - $C_6$  alkyl part of alkylHet and alkylaryl groups), when used herein, includes (e.g. methyl, ethyl, propyl, butyl, pentyl and hexyl groups). Unless otherwise specified,  $C_1$ - $C_6$  alkyl groups may, when there is a sufficient number of carbon atoms, be linear or branched or be saturated or unsaturated.

[0011] As defined herein, the term "halo" includes fluoro, chloro, bromo and iodo.

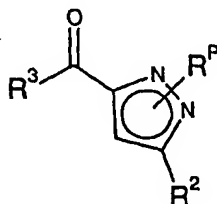
[0012] Suitable electron donating groups, EDGs, for use herein include: trialkylsilyl.

[0013] Suitable electron withdrawing groups, EWGs, for use herein include: *tert*-butoxycarbonyl and trifluoroacetamide.

[0014] Suitable bases for use herein preferably include: tertiary amines, such as triethylamine and di-*iso*-propylethylamine, 1,5-diazabicyclo[4.3.0]non-5-ene, 1,8-diazabicyclo[5.4.0]undec-7-ene, 1,4-diazabicyclo[2.2.2]octane, imidazole; substituted pyridines, such as 4-(dimethylamino)pyridine; benzofused pyridines, such as quinoline and isoquinoline; hindered metal alkoxides; hindered metal aryloxides; metal carbonates and bicarbonates.

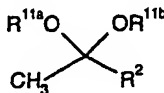
[0015] Compounds of formulae (II) may contain one or more asymmetric carbon atoms and may therefore exhibit optical and/or diastereoisomerism. The process of the invention thus also relates to the formation of stereoisomers of compounds of formulae II and mixtures thereof. Stereoisomers may be separated using conventional techniques, e.g. chromatography or fractional crystallisation. The various stereoisomers may be isolated by separation of a racemic or other mixture of the compounds using conventional, e.g. fractional crystallisation or HPLC, techniques. Alternatively the desired optical isomers may be made by reaction of the appropriate optically active starting materials under conditions which will not cause racemisation or epimerisation, or by derivatisation, for example with a homochiral acid followed by separation of the diastereomeric esters by conventional means (e.g. HPLC, crystallisation, chromatography over silica or, for example, *via* classical resolution with a homochiral acid salt). The formation of all stereoisomers is included within the scope of the invention.

[0016] In a preferred aspect of the present invention there is provided a process for the production of pyrazole compounds of general formula (II):



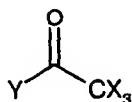
II

wherein  $R^P$  represents: H or  $R^1$ ;  $R^1$  represents  $C_1$ - $C_4$  alkyl, which alkyl group is optionally substituted as hereinbefore described;  $R^2$  represents  $C_1$ - $C_4$  alkyl;  $R^3$  represents  $C_1$ - $C_3$  alkoxy, which process comprises the reaction of a compound (III),



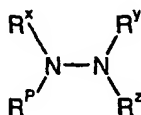
III

wherein  $R^{11a}$  and  $R^{11b}$  independently represent:  $C_1$ - $C_4$  alkyl and  $R^2$  represents  $C_1$ - $C_4$  alkyl, with an acylating agent of the formula (IV) in the presence of a base, preferably pyridine:



IV

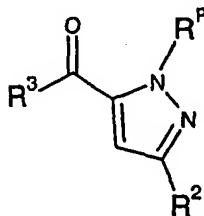
wherein X represents Cl or F and Y represents Cl, F or C(O)CX<sub>3</sub>, optionally in the presence of an activating agent, followed by reaction with a hydrazine compound of general formula (V) in the same vessel:



V

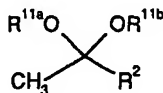
wherein R<sup>p</sup> represents H or R<sup>1</sup> wherein R<sup>1</sup> is as defined hereinbefore, and R<sup>x</sup>, R<sup>y</sup> and R<sup>z</sup> are independently selected from H or electron donating groups or electron withdrawing groups wherein said electron withdrawing or donating groups are labile under the conditions of the reaction.

**[0017]** In a further preferred aspect of the present invention there is provided a process for the production of pyrazole compounds of general formula (IIA):



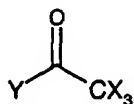
IIA

wherein R<sup>p</sup> represents: H or R<sup>1</sup>; R<sup>1</sup> represents C<sub>1</sub>-C<sub>4</sub> alkyl, which alkyl group is optionally substituted as hereinbefore described; R<sup>2</sup> represents C<sub>1</sub>-C<sub>4</sub> alkyl; R<sup>3</sup> represents C<sub>1</sub>-C<sub>3</sub> alkoxy, which process comprises the reaction of a compound (III):



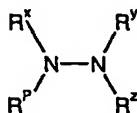
III

wherein R<sup>11a</sup> and R<sup>11b</sup> independently represent: C<sub>1</sub>-C<sub>4</sub> alkyl and R<sup>2</sup> represents C<sub>1</sub>-C<sub>4</sub> alkyl, with an acylating agent in the presence of a base, preferably pyridine, of the formula (IV):



IV

wherein X represents Cl or F and Y represents Cl, F or C(O)CX<sub>3</sub>, optionally in the presence of an activating agent, followed by reaction with a hydrazine compound of general formula (V):



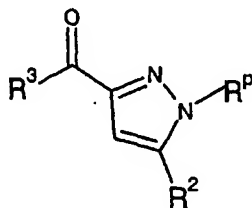
V

wherein R<sup>p</sup> represents H or R<sup>1</sup> wherein R<sup>1</sup> is as defined hereinbefore, and wherein R<sup>x</sup> = EWG, R<sup>y</sup> and R<sup>z</sup> = H; or

R<sup>x</sup> and R<sup>y</sup> = H and R<sup>z</sup> = EDG; or  
 R<sup>x</sup> = R<sup>y</sup> = R<sup>z</sup> = H when R<sup>1</sup> = EWG; or  
 R<sup>x</sup> = EWG, R<sup>y</sup> = H and R<sup>z</sup> = EDG

wherein EWG is a tri(C<sub>1</sub>-C<sub>2</sub>)alkylsilyl group and wherein EDG is selected from *tert*-butoxycarbonyl and trifluoroacetamide.

**[0018]** In yet another preferred aspect of the present invention there is provided a process for the production of pyrazole compounds of general formula (IIB):

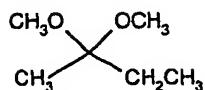


IIB

wherein R<sup>p</sup> represents H or R<sup>1</sup> wherein R<sup>1</sup> represents C<sub>1</sub>-C<sub>4</sub> alkyl or C<sub>1</sub>-C<sub>4</sub> alkyl(C<sub>1</sub>-C<sub>2</sub> alkoxy);

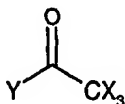
R<sup>2</sup> represents C<sub>1</sub>-C<sub>4</sub> alkyl; and  
 R<sup>3</sup> represents C<sub>1</sub>-C<sub>3</sub> alkoxy,

which process comprises the reaction of a compound (III):



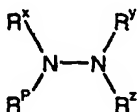
III

with an acylating agent, in the presence of pyridine, of the formula (IV):



IV

wherein X and Y independently represent Cl or F, optionally in the presence of an activating agent, followed, after between about 8 and 24 hours by addition of a hydrazine compound of general formula (V):



V

wherein when  $\text{R}^p$  represents H, then

$\text{R}^x = \text{EWG}$ ,  $\text{R}^y$  and  $\text{R}^z = \text{H}$ ; or

$\text{R}^x$  and  $\text{R}^y = \text{H}$  and  $\text{R}^z = \text{EDG}$ ; or

$\text{R}^x = \text{R}^y = \text{R}^z = \text{H}$  when  $\text{R}^1 = \text{EWG}$ ; or

$\text{R}^x = \text{EWG}$ ,  $\text{R}^y = \text{H}$  and  $\text{R}^z = \text{EDG}$  wherein EDG is a tri( $\text{C}_1\text{-C}_2$ )alkylsilyl group and wherein EWG is selected from *tert*-butyloxycarbonyl and trifluoroacetamide.

and wherein when  $\text{R}^p$  represents  $\text{R}^1$  wherein  $\text{R}^1$  is as defined hereinbefore, then

$\text{R}^x = \text{H}$ ,  $\text{R}^y = \text{H}$  and  $\text{R}^z = \text{EWG}$ ; or

$\text{R}^x = \text{EDG}$ ,  $\text{R}^y = \text{H}$  and  $\text{R}^z = \text{H}$ ; or

$\text{R}^x = \text{R}^y = \text{R}^z = \text{H}$  when  $\text{R}^1 = \text{EDG}$ ; or

$\text{R}^x = \text{EDG}$ ,  $\text{R}^y = \text{H}$  and  $\text{R}^z = \text{EWG}$  wherein EDG is a tri( $\text{C}_1\text{-C}_2$ )alkylsilyl group and wherein EWG is selected from *tert*-butyloxycarbonyl and trifluoroacetamide.

**[0019]** The process of the invention may be carried out in accordance with reaction conditions known to those skilled in the art

**[0020]** The process according to the present invention requires an activating agent to "activate" the compound of general formula (III) and at least one equivalent of an acylating agent of general formula (IV) to "react" with the activated compound generated from the compound of general formula (III). Any suitable agent capable of activating the compound of general formula (III), may be used in conjunction with at least one equivalent of acylating agent of general formula (IV) according to the process of the present invention.

**[0021]** The activating agent should be capable of converting an acetal to an enol ether under the basic reaction conditions. Suitable activating agents could be, trialkylsilyl halides, trialkylsilyl trifluoromethanesulfonates, oxalyl halides.

des, 2-(trifluoroacetoxy)pyridine, 1-(trifluoroacetyl)imidazole, trifluoroacetyl chloride, trifluoroacetic anhydride, tribromoacetyl chloride and trichloroacetyl chloride.

[0022] More preferable activators are 2-(trifluoroacetoxy)pyridine, 1-(trifluoroacetyl)imidazole, trifluoroacetyl chloride, trifluoroacetic anhydride, tribromoacetyl chloride and trichloroacetyl chloride.

5 [0023] In a preferred process according to the present invention the acylating agent of general formula (IV) can be used as both the activating agent and the acylating agent.

The acylating agent (IV) must be capable of acylating an enol ether to afford the key enone intermediate. Suitable reagents include be 2-(trifluoroacetoxy)pyridine, 1-(trifluoroacetyl)imidazole, trifluoroacetyl chloride, trifluoroacetic anhydride, tribromoacetyl chloride and trichloroacetyl chloride. Preferred for use herein are trifluoroacetic anhydride and trichloroacetyl chloride. Even more preferred for use herein is trichloroacetylchloride.

10 [0024] Thus according to a yet further aspect of the present invention there is provided a process for the production of pyrazole compounds of general formula (II), as defined hereinbefore, which process comprises the reaction of a compound (III), as defined herein before, with at least one equivalent, more preferably at least two equivalents of an acylating agent of the formula (IV), as defined hereinbefore, optionally in the presence of an activating agent, followed by reaction with a hydrazine compound of general formula (V), as defined hereinbefore.

15 [0025] The reaction between the compound of the general formula (III) and the activating agent (and/or the acylating agent of general formula (IV), for reactions wherein the activating and acylating agents are the same) according to the process of the present invention may be carried out in an appropriate organic solvent system, which solvent system should not significantly react chemically with, or significantly give rise to stereochemical changes in, the reactants or product once formed, or significantly give rise to other side reactions. Suitable solvents include: halogenated hydrocarbons (such as chloroform, dichloromethane and 1,2-dichloroethane), ethers (such as tetrahydrofuran, 1,4-dioxan, diethyl ether and *tert*-butyl methyl ether), aromatic hydrocarbons (such as toluene, xylenes and chlorobenzene) and alkyl acetates (such as ethyl acetate) and mixtures thereof. A preferred solvent is dichloromethane.

25 [0026] The reaction between the compound of the general formula (III) and the activating agent (and/or acylating agent of general formula (IV)) according to the process of the present invention may be carried at from 0°C to about room temperature, and, preferably, in an inert atmosphere (i.e. in the presence of an inert gas, such as nitrogen or argon).

[0027] Following the activation and acylation of the compound of general formula (III) the hydrazine compound of general formula (V) is added directly to the reaction mixture (of the activating agent (and/or acylating agent of general formula (IV)) and the compound of general formula (III)) *in situ* to provide a compound of general formula (II) according to the process of the present invention. The hydrazine compound (V) may be added portionwise, dropwise, in solution or neat. Typically the hydrazine compound is added in water and/or a suitable organic solvent (e.g. alcohols such as methanol, ethanol or *iso*-propanol, to provide a compound of general formula (II) wherein R<sup>3</sup> is C<sub>1</sub> to C<sub>6</sub> alkoxy), or aqueous ammonia (to provide a compound of general formula (II) wherein R<sup>3</sup> is NR<sup>4</sup>R<sup>5</sup>) or mixtures thereof, followed by removal of the original reaction solvent (e.g. dichloromethane) and heat treatment.

35 [0028] In reactions with hydrazine compounds of general formula (V) wherein R<sup>P</sup> = R<sup>1</sup> and wherein R<sup>1</sup> is an acid labile group such as a *tert*-butoxycarbonyl group, the pH of the reaction mixture may be adjusted to between about pH 1.5 and about pH 3 and preferably to about pH 2 following addition of the hydrazine compound.

40 [0029] In reactions with hydrazine compounds of general formula (V) wherein R<sup>P</sup> = H, it is not necessary to adjust the pH of the reaction mixture following addition of the hydrazine compound.

[0030] In a yet further aspect of the process according to the present invention compounds of general formula (II) wherein R<sup>3</sup> = OH may be prepared either via conversion of pyrazole compounds of general formula (II) wherein R<sup>3</sup> = alkoxy as obtained according to the process described hereinbefore, or, alternatively via an *in situ* conversion (of the ester to the acid) wherein the pH of the reaction mixture is raised to greater than about pH 8, via addition of a suitable base such as for example NaOH.

45 [0031] In a yet further aspect of the process according to the present invention compounds of general formula (II) prepared as outlined hereinbefore via (III), (IV) and (V) using aqueous ammonia as solvent i.e. a compound of general formula (II) wherein R<sup>3</sup> is NR<sup>4</sup>R<sup>5</sup> may be converted directly to provide a compound of the general formula B (as illustrated in Scheme 1).

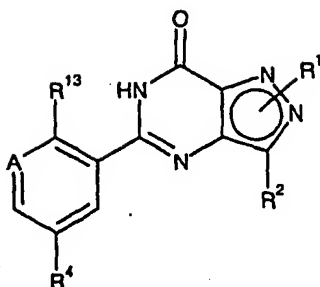
50 [0032] Appropriate reaction times and reaction temperatures depend upon the solvent system that is employed, as well as the compound that is to be formed, but these may be determined routinely by the skilled person.

[0033] Compounds of general formula (III) when not commercially available may be prepared by known techniques as detailed in the preparations section herein.

55 [0034] Compounds of general formulae (IV) and (V), and derivatives thereof, when not commercially available or not subsequently described, may be obtained by conventional synthetic procedures or by analogy with the processes described herein, in accordance with standard techniques, from readily available starting materials using appropriate reagents and reaction conditions.

[0035] According to a further aspect of the present invention the compounds of general formula general formula I:





(I)

or a pharmaceutically or veterinarily acceptable salt thereof, or a pharmaceutically or veterinarily acceptable solvate of either entity, wherein:

A is CH or N;

R<sup>1</sup> is H, C<sub>1</sub> to C<sub>6</sub> alkyl, C<sub>3</sub> to C<sub>6</sub> alkenyl, C<sub>3</sub> to C<sub>6</sub> cycloalkyl, C<sub>3</sub> to C<sub>6</sub> cycloalkenyl, or C<sub>1</sub>-C<sub>3</sub> perfluoroalkyl, wherein said alkyl group may be branched or straight chain and wherein said alkyl, alkenyl, cycloalkyl or perfluoroalkyl group is optionally substituted by: one or more substituents selected from: hydroxy; C<sub>1</sub> to C<sub>4</sub> alkoxy; C<sub>3</sub> to C<sub>6</sub> cycloalkyl; C<sub>1</sub>-C<sub>3</sub> perfluoroalkyl; phenyl substituted with one or more substituents selected from C<sub>1</sub> to C<sub>3</sub> alkyl, C<sub>1</sub> to C<sub>4</sub> alkoxy, C<sub>1</sub> to C<sub>4</sub> haloalkyl or C<sub>1</sub> to C<sub>4</sub> haloalkoxy wherein said haloalkyl and haloalkoxy groups contain one or more halo atoms, halo, CN, NO<sub>2</sub>, NHR<sup>11</sup>, NHSO<sub>2</sub>R<sup>12</sup>, SO<sub>2</sub>R<sup>12</sup>, SO<sub>2</sub>NHR<sup>11</sup>, COR<sup>11</sup>, CO<sub>2</sub>R<sup>11</sup> wherein R<sup>11</sup> is H, C<sub>1</sub> to C<sub>4</sub> alkyl, C<sub>2</sub> to C<sub>4</sub> alkenyl, C<sub>1</sub> to C<sub>4</sub> alkanoyl, C<sub>1</sub> to C<sub>4</sub> haloalkyl or C<sub>1</sub> to C<sub>4</sub> haloalkoxy and wherein R<sup>12</sup> is C<sub>1</sub> to C<sub>4</sub> alkyl, C<sub>2</sub> to C<sub>4</sub> alkenyl, C<sub>1</sub> to C<sub>4</sub> alkanoyl, C<sub>1</sub> to C<sub>4</sub> haloalkyl or C<sub>1</sub> to C<sub>4</sub> haloalkoxy; NR<sup>7</sup>R<sup>8</sup>, CONR<sup>7</sup>R<sup>8</sup> or NR<sup>7</sup>COR<sup>11</sup> wherein R<sup>7</sup> and R<sup>8</sup> are each independently selected from H, C<sub>1</sub> to C<sub>4</sub> alkyl, C<sub>2</sub> to C<sub>4</sub> alkenyl, C<sub>1</sub> to C<sub>4</sub> alkoxy, CO<sub>2</sub>R<sup>9</sup>, SO<sub>2</sub>R<sup>9</sup> wherein said alkyl, alkenyl or alkoxy groups are optionally substituted by NR<sup>5</sup>R<sup>6</sup>, C<sub>1</sub> to C<sub>4</sub> haloalkyl or C<sub>1</sub> to C<sub>4</sub> haloalkoxy and wherein R<sup>9</sup> is H, hydroxy C<sub>2</sub> to C<sub>3</sub> alkyl, C<sub>1</sub> to C<sub>4</sub> alkanoyl or C<sub>1</sub> to C<sub>4</sub> alkyl which is optionally substituted with phenyl wherein said phenyl group is optionally substituted by one or more substituents selected from C<sub>1</sub> to C<sub>4</sub> alkyl optionally substituted by C<sub>1</sub> to C<sub>4</sub> haloalkyl or C<sub>1</sub> to C<sub>4</sub> haloalkoxy, C<sub>1</sub> to C<sub>4</sub> alkoxy, halo, CN, NO<sub>2</sub>, NHR<sup>11</sup>, NHSO<sub>2</sub>R<sup>12</sup>, SO<sub>2</sub>R<sup>12</sup>, SO<sub>2</sub>NHR<sup>11</sup>, COR<sup>11</sup> or CO<sub>2</sub>R<sup>11</sup>; Het<sup>1</sup>; Het<sup>2</sup> or Het<sup>3</sup>; or R<sup>1</sup> is Het<sup>4</sup> or phenyl wherein said phenyl group is optionally substituted by one or more substituents selected from C<sub>1</sub> to C<sub>4</sub> alkyl, C<sub>2</sub> to C<sub>4</sub> alkenyl, C<sub>1</sub> to C<sub>4</sub> alkoxy, halo, CN, CF<sub>3</sub>, OCF<sub>3</sub>, NO<sub>2</sub>, NHR<sup>11</sup>, NHSO<sub>2</sub>R<sup>12</sup>, SO<sub>2</sub>R<sup>12</sup>, SO<sub>2</sub>NHR<sup>11</sup>, COR<sup>11</sup>, CO<sub>2</sub>R<sup>11</sup>;

R<sup>2</sup> is H, C<sub>1</sub> to C<sub>6</sub> alkyl, C<sub>3</sub> to C<sub>6</sub> alkenyl or (CH<sub>2</sub>)<sub>n</sub>(C<sub>3</sub> to C<sub>6</sub> cycloalkyl) wherein n is 0, 1 or 2 and wherein said alkyl or alkenyl group is optionally substituted with one or more fluoro substituents;

R<sup>13</sup> is OR<sup>3</sup> or NR<sup>5</sup>R<sup>6</sup>;

R<sup>3</sup> is C<sub>1</sub> to C<sub>6</sub> alkyl, C<sub>3</sub>-C<sub>6</sub> alkenyl, C<sub>3</sub>-C<sub>6</sub> alkynyl, C<sub>3</sub>-C<sub>7</sub> cycloalkyl, C<sub>1</sub>-C<sub>6</sub> perfluoroalkyl or (C<sub>3</sub>-C<sub>6</sub> cycloalkyl)C<sub>1</sub>-C<sub>6</sub> alkyl optionally substituted with one or two substituents selected from C<sub>3</sub> to C<sub>5</sub> cycloalkyl, hydroxy, C<sub>1</sub> to C<sub>4</sub> alkoxy, C<sub>3</sub>-C<sub>6</sub> alkenyl, C<sub>3</sub>-C<sub>6</sub> alkynyl, benzyloxy, NR<sup>5</sup>R<sup>6</sup>, phenyl, Het<sup>1</sup>, Het<sup>2</sup>, Het<sup>3</sup> or Het<sup>4</sup> wherein the C<sub>1</sub> to C<sub>6</sub> alkyl and C<sub>1</sub> to C<sub>4</sub> alkoxy groups may optionally be terminated by a haloalkyl group such as CF<sub>3</sub>; C<sub>3</sub> to C<sub>6</sub> cycloalkyl; Het<sup>1</sup>, Het<sup>2</sup>, Het<sup>3</sup> or Het<sup>4</sup>;

R<sup>4</sup> is C<sub>1</sub>-C<sub>4</sub> alkyl optionally substituted with OH, NR<sup>5</sup>R<sup>6</sup>, CN, CONR<sup>5</sup>R<sup>6</sup> or CO<sub>2</sub>R<sup>7</sup>; C<sub>2</sub>-C<sub>4</sub> alkenyl optionally substituted with CN, CONR<sup>5</sup>R<sup>6</sup> or CO<sub>2</sub>R<sup>7</sup>; C<sub>2</sub>-C<sub>4</sub> alkanoyl optionally substituted with NR<sup>5</sup>R<sup>6</sup>; hydroxy C<sub>2</sub>-C<sub>4</sub> alkyl optionally substituted with NR<sup>5</sup>R<sup>6</sup>; (C<sub>2</sub>-C<sub>3</sub> alkoxy)C<sub>1</sub>-C<sub>2</sub> alkyl optionally substituted with OH or NR<sup>5</sup>R<sup>6</sup>; CONR<sup>5</sup>R<sup>6</sup>; CO<sub>2</sub>R<sup>7</sup>; halo; NR<sup>5</sup>R<sup>6</sup>; NHSO<sub>2</sub>NR<sup>5</sup>R<sup>6</sup>; NHSO<sub>2</sub>R<sup>8</sup>; or phenyl or heterocyclyl either of which is optionally substituted with methyl; or R<sup>4</sup> is a pyrrolidinylsulphonyl, piperidinylsulphonyl, morpholinylsulphonyl, or piperazin-1-ylsulphonyl group having a substituent, R<sup>10</sup> at the 4-position of the piperazinyl group wherein said piperazinyl group is optionally substituted with one or two C<sub>1</sub> to C<sub>4</sub> alkyl, C<sub>1</sub> to C<sub>3</sub> alkoxy, NR<sup>7</sup>R<sup>8</sup> or CONR<sup>7</sup>R<sup>8</sup> groups and is optionally in the form of its 4-N-oxide;

R<sup>5</sup> and R<sup>6</sup> are each independently selected from H and C<sub>1</sub> to C<sub>4</sub> alkyl optionally substituted with C<sub>3</sub> to C<sub>5</sub> cycloalkyl

or C<sub>1</sub> to C<sub>4</sub> alkoxy, or, together with the nitrogen atom to which they are attached, form an azetidiny, pyrrolidinyl, piperidinyl, morpholinyl, 4-(NR<sup>9</sup>)-piperazinyl or imidazolyl group wherein said group is optionally substituted with methyl or hydroxy;

R<sup>10</sup> is H; C<sub>1</sub> to C<sub>6</sub> alkyl, (C<sub>1</sub>-C<sub>3</sub> alkoxy) C<sub>2</sub>-C<sub>6</sub> alkyl, hydroxy C<sub>2</sub>-C<sub>6</sub> alkyl, (R<sup>7</sup>R<sup>8</sup>N)C<sub>2</sub>-C<sub>6</sub> alkyl, (R<sup>7</sup>R<sup>8</sup>NCO)C<sub>1</sub>-C<sub>6</sub> alkyl, CONR<sup>7</sup>R<sup>8</sup>, CSNR<sup>7</sup>R<sup>8</sup> or C(NH)NR<sup>7</sup>R<sup>8</sup> optionally substituted with one or two substituents selected from hydroxy, NR<sup>5</sup>R<sup>6</sup>, CONR<sup>5</sup>R<sup>6</sup>, phenyl optionally substituted with C<sub>1</sub> to C<sub>4</sub> alkyl or C<sub>1</sub> to C<sub>4</sub> alkoxy; C<sub>2</sub> to C<sub>6</sub> alkenyl or Het<sup>4</sup>;

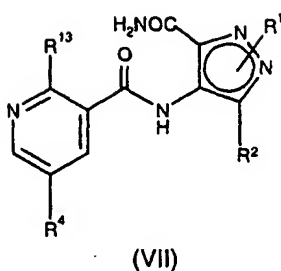
Het<sup>1</sup> is an N-linked 4-, 5- or 6-membered nitrogen-containing heterocyclic group optionally containing one or more further heteroatoms selected from S, N or O;

Het<sup>2</sup> is a C-linked 5-membered heterocyclic group containing an O, S or N heteroatom optionally containing one or more heteroatoms selected from O or S;

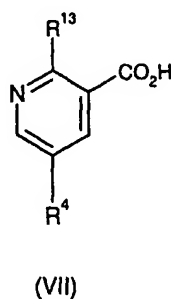
Het<sup>3</sup> is a C-linked 6-membered heterocyclic group containing an O or S heteroatom optionally containing one or more heteroatoms selected from O, S or N or Het<sup>3</sup> is a C-linked 6-membered heterocyclic group containing three N heteroatoms;

Het<sup>4</sup> is a C-linked 4-, 5- or 6-membered heterocyclic group containing one, two or three heteroatoms selected from S, O or N; and wherein any of said heterocyclic groups Het<sup>1</sup>, Het<sup>2</sup>, Het<sup>3</sup> or Het<sup>4</sup> may be saturated, partially unsaturated or aromatic and wherein any of said heterocyclic groups may be optionally substituted with one or more substituents selected from C<sub>1</sub> to C<sub>4</sub> alkyl, C<sub>2</sub> to C<sub>4</sub> alkenyl, C<sub>1</sub> to C<sub>4</sub> alkoxy, halo, CO<sub>2</sub>R<sup>11</sup>, COR<sup>11</sup>, SO<sub>2</sub>R<sup>12</sup> or NHR<sup>11</sup> and/or wherein any of said heterocyclic groups is benzo-fused;

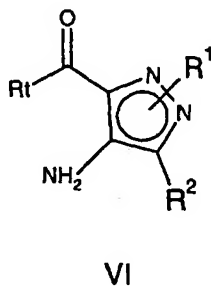
or wherein when R<sup>13</sup> represents OR<sup>3</sup> or R<sup>3</sup>NR<sup>5</sup>; R<sup>1</sup> represents Het, alkylHet, aryl or alkylaryl, which latter five groups are all optionally substituted and/or terminated with one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, halo(C<sub>1</sub>-C<sub>6</sub> alkyl), OR<sup>6</sup>, OC(O)R<sup>7</sup>, C(O)R<sup>8</sup>, C(O)OR<sup>9</sup>, C(O)NR<sup>10</sup>R<sup>11</sup>, NR<sup>12</sup>R<sup>13</sup> and SO<sub>2</sub>NR<sup>14</sup>R<sup>15</sup>; R<sup>2</sup> represents H, halo, cyano, nitro, OR<sup>6</sup>, OC(O)R<sup>7</sup>, C(O)R<sup>8</sup>, C(O)OR<sup>9</sup>, C(O)NR<sup>10</sup>R<sup>11</sup>, NR<sup>12</sup>R<sup>13</sup>, SO<sub>2</sub>NR<sup>14</sup>R<sup>15</sup>, C<sub>1</sub>-C<sub>6</sub> alkyl, Het, alkylHet, aryl or alkylaryl, which latter five groups are all optionally substituted and/or terminated with one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, halo(C<sub>1</sub>-C<sub>6</sub> alkyl), OR<sup>6</sup>, OC(O)R<sup>7</sup>, C(O)R<sup>8</sup>, C(O)OR<sup>9</sup>, C(O)NR<sup>10</sup>R<sup>11</sup>, NR<sup>12</sup>R<sup>13</sup> and SO<sub>2</sub>NR<sup>14</sup>R<sup>15</sup>; R<sup>3</sup> represents H, C<sub>1</sub>-C<sub>6</sub> alkyl, alkylHet or alkylaryl, which latter three groups are all optionally substituted and/or terminated with one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, halo(C<sub>1</sub>-C<sub>6</sub> alkyl), OR<sup>6</sup>, OC(O)R<sup>7</sup>, C(O)R<sup>8</sup>, C(O)OR<sup>9</sup>, C(O)NR<sup>10</sup>R<sup>11</sup>, NR<sup>12</sup>R<sup>13</sup> and SO<sub>2</sub>NR<sup>14</sup>R<sup>15</sup>; R<sup>4</sup> represents H, halo, cyano, nitro, halo(C<sub>1</sub>-C<sub>6</sub> alkyl), OR<sup>6</sup>, OC(O)R<sup>7</sup>, C(O)R<sup>8</sup>, C(O)OR<sup>9</sup>, C(O)NR<sup>10</sup>R<sup>11</sup>, NR<sup>12</sup>R<sup>13</sup>, NR<sup>16</sup>Y(O)R<sup>17</sup>, SOR<sup>18</sup>, SO<sub>2</sub>R<sup>19</sup>R<sup>20</sup>, C(O)AZ, C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>2</sub>-C<sub>6</sub> alkenyl, C<sub>2</sub>-C<sub>6</sub> alkynyl, Het, alkylHet, aryl, alkylaryl, which latter seven groups are all optionally substituted and/or terminated with one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, halo(C<sub>1</sub>-C<sub>6</sub> alkyl), OR<sup>6</sup>, OC(O)R<sup>7</sup>, C(O)R<sup>8</sup>, C(O)OR<sup>9</sup>, C(O)NR<sup>10</sup>R<sup>11</sup>, NR<sup>12</sup>R<sup>13</sup> and SO<sub>2</sub>NR<sup>14</sup>R<sup>15</sup>; Y represents C or S(O), wherein one of R<sup>16</sup> and R<sup>17</sup> is not present when Y is S(O); A represents C<sub>2</sub>-C<sub>6</sub> alkylene; Z represents OR<sup>6</sup>, halo, Het or aryl, which latter two groups are both optionally substituted with one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, halo(C<sub>1</sub>-C<sub>6</sub> alkyl), OR<sup>6</sup>, OC(O)R<sup>7</sup>, C(O)R<sup>8</sup>, C(O)OR<sup>9</sup>, C(O)NR<sup>10</sup>R<sup>11</sup>, NR<sup>12</sup>R<sup>13</sup> and SO<sub>2</sub>NR<sup>14</sup>R<sup>15</sup>; R<sup>5</sup>, R<sup>6</sup>, R<sup>7</sup>, R<sup>8</sup>, R<sup>9</sup>, R<sup>18</sup>, R<sup>19</sup> and R<sup>20</sup> independently represent H or C<sub>1</sub>-C<sub>6</sub> alkyl; R<sup>10</sup> and R<sup>11</sup> independently represent H or C<sub>1</sub>-C<sub>6</sub> alkyl, which latter group is optionally substituted and/or terminated with one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, halo(C<sub>1</sub>-C<sub>6</sub> alkyl), OR<sup>6</sup>, OC(O)R<sup>7</sup>, C(O)R<sup>8</sup>, C(O)OR<sup>9</sup>, C(O)NR<sup>10</sup>R<sup>11</sup>, NR<sup>12</sup>R<sup>13</sup> and SO<sub>2</sub>NR<sup>14</sup>R<sup>15</sup> or Het or aryl optionally substituted with one or more of said latter eleven groups or one of R<sup>10</sup> and R<sup>11</sup> may be C<sub>1</sub>-C<sub>6</sub> alkoxy, amino or Het, which latter two groups are both optionally substituted with C<sub>1</sub>-C<sub>6</sub> alkyl; R<sup>12</sup> and R<sup>13</sup> independently represent H or C<sub>1</sub>-C<sub>6</sub> alkyl or one of R<sup>12</sup> or R<sup>13</sup> may be C(O)-C<sub>1</sub>-C<sub>6</sub> alkyl or C(O)Het in which Het is optionally substituted with C<sub>1</sub>-C<sub>6</sub> alkyl; R<sup>14</sup> and R<sup>15</sup> independently represent H or C<sub>1</sub>-C<sub>6</sub> alkyl or R<sup>14</sup> and R<sup>15</sup>, together with the nitrogen atom to which they are bound, form a heterocyclic ring; R<sup>16</sup> and R<sup>17</sup> independently represent H or C<sub>1</sub>-C<sub>6</sub> alkyl or one of R<sup>16</sup> and R<sup>17</sup> may be Het or aryl, which latter two groups are both optionally substituted with C<sub>1</sub>-C<sub>6</sub> alkyl; Het represents an optionally substituted four to twelve membered heterocyclic group, which may be aromatic or non-aromatic, which may contain one or more double bonds, which may be mono- or bicyclic and which contains one or more heteroatoms selected from N, S and O; wherein said compounds may be prepared from compounds of general formula (VIII):



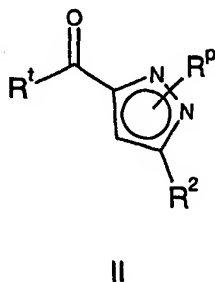
wherein  $R^1$ ,  $R^2$ ,  $R^4$  and  $R^{13}$  are as defined hereinbefore wherein said compound of general formula (VIII) is prepared from the reaction of a compound of general formula (VII):



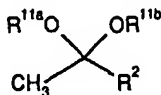
wherein  $R^4$  and  $R^{13}$  are as defined herein before via coupling with a compound of general formula (VI):



wherein  $R^1$  and  $R^2$  are as defined hereinbefore and wherein  $R^1$  is  $NR^pR^q$  wherein  $R^p$  and  $R^q$  independently represent H or  $C_1$ - $C_6$  alkyl and wherein said compound of general formula (VI) is prepared by nitration and hydrogenation of a compound of general formula (II):

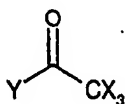


wherein  $R^1$  and  $R^2$  are as defined hereinbefore and wherein  $R^P$  represents  $R^1$  as defined hereinbefore characterised in that said compound of general formula (II) is prepared from the reaction of a compound of general formula (III),



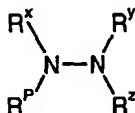
III

wherein  $R^{11a}$  and  $R^{11b}$  independently represent  $C_1$ - $C_6$  alkyl,  $R^2$  is as defined herein before, with an acylating agent of the formula (IV) in the presence of a base,



IV

wherein X represents halogen independently selected from Cl, F or Br, and Y represents halogen or  $OR^{12}$  wherein  $R^{12}$  is:  $C_1$ - $C_6$  alkyl;  $C(O)CX_3$ ; Het; or  $C_0$ - $C_6$  alkyl(Het) wherein Het is pyridine or imidazole, followed by *in situ* reaction with a hydrazine compound of general formula (V)



V

$R^P$  represents H or  $R^1$  wherein  $R^1$  is as defined hereinbefore, and  $R^x$ ,  $R^y$  and  $R^z$  are independently selected from H or electron donating groups (EDG) or electron withdrawing groups (EWG) wherein said electron withdrawing or donating groups are labile under the conditions of the reaction,

wherein said process for the preparation of compound (II) is optionally carried out in the presence of an activating agent.

**[0036]** According to a preferred process of the present invention compounds of formula (I) are prepared from compounds of formula (II) wherein  $R^1$  represents: H,  $C_1$ - $C_6$  alkyl;  $C_1$ - $C_6$  alkoxy;  $C_3$ - $C_6$  cycloalkyl;  $C_1$ - $C_6$  alkyl( $C_1$ - $C_6$  alkoxy), Het,  $C_1$ - $C_6$  alkylHet, aryl or  $C_1$ - $C_6$  alkylaryl, which latter eight groups are all optionally substituted (and/or, in the case of  $C_1$ - $C_6$  alkyl, optionally terminated) by one or more substituents selected from halo, cyano, nitro,  $C_1$ - $C_6$  alkyl,  $C(O)NR^4R^5$ ,  $C(O)R^6$ ,  $C(O)OR^7$ ,  $OR^8$ ,  $NR^{9a}R^{9b}$  and  $SO_2NR^{10a}R^{10b}$ ;

$R^2$  represents  $C_1$ - $C_6$  alkyl;  $C_1$ - $C_6$  alkoxy;  $C_3$ - $C_6$  cycloalkyl;  $C_1$ - $C_6$  alkyl( $C_1$ - $C_6$  alkoxy), Het,  $C_1$ - $C_6$  alkylHet, aryl or  $C_1$ - $C_6$  alkylaryl, which latter eight groups are all optionally substituted (and/or, in the case of  $C_1$ - $C_6$  alkyl, optionally terminated) by one or more substituents selected from halo, cyano, nitro,  $C_1$ - $C_6$  alkyl,  $C(O)NR^4R^5$ ,  $C(O)R^6$ ,  $C(O)OR^7$ ,  $OR^8$ ,  $NR^{9a}R^{9b}$  and  $SO_2NR^{10a}R^{10b}$ ;

$R^1$  represents:  $NR^PR^Q$ ;

$R^P$ ,  $R^Q$ ,  $R^6$ ,  $R^7$ ,  $R^8$ ,  $R^{10a}$  and  $R^{10b}$  independently represent H or  $C_1$ - $C_6$  alkyl;

$R^{9a}$  and  $R^{9b}$  either independently represent, H or  $C_1$ - $C_6$  alkyl or, together with the nitrogen atom to which they are attached, represent azetidiny, pyrrolidiny or piperidinyl;

R<sup>4</sup> is C<sub>1</sub>-C<sub>4</sub> alkyl optionally substituted with OH, NR<sup>5</sup>R<sup>6</sup>, CN, CONR<sup>5</sup>R<sup>6</sup> or CO<sub>2</sub>R<sup>7</sup>; or R<sup>4</sup> is CO<sub>2</sub>R<sup>7</sup>; or R<sup>4</sup> is a pyrrolidinylsulphonyl, piperidinylsulphonyl, morpholinylsulphonyl, or piperazin-1-ylsulphonyl group having a substituent, R<sup>10</sup> at the 4-position of the piperazinyl group wherein said piperazinyl group is optionally substituted with one or two C<sub>1</sub> to C<sub>4</sub> alkyl, C<sub>1</sub> to C<sub>3</sub> alkoxy, NR<sup>7</sup>R<sup>8</sup> or CONR<sup>7</sup>R<sup>8</sup> groups and is optionally in the form of its 4-N-oxide; R<sup>10</sup> represents H or C<sub>1</sub>-C<sub>6</sub> alkyl.

[0037] Highly preferred herein is a process for the preparation of compounds of general formula (I) from compounds of general formula (II) wherein R<sup>1</sup> represents C<sub>1-4</sub> alkyl, which alkyl group is optionally interrupted by an oxygen atom, and/or is optionally terminated by a Het group (such as a pyridinyl group);

R<sup>2</sup> represents C<sub>1-4</sub> alkyl;

R<sup>3</sup> represents C<sub>1-5</sub> alkyl, which alkyl group is optionally interrupted by an oxygen atom;

R<sup>4</sup> is CO<sub>2</sub>R<sup>7</sup>; or R<sup>4</sup> is a morpholinylsulphonyl, or piperazin-1-ylsulphonyl group having a substituent, R<sup>10</sup> at the 4-position of the piperazinyl group wherein R<sup>10</sup> represents H, methyl or ethyl.

[0038] More preferred compounds of formulae I, IA and IB prepared according to a process of the invention include those in which:

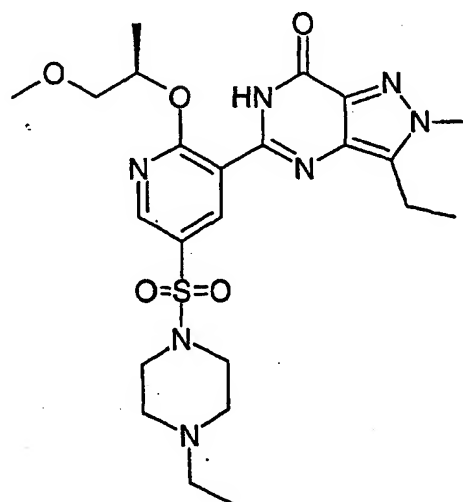
R<sup>1</sup> represents linear C<sub>1-3</sub> alkyl, which alkyl group is optionally interrupted by an oxygen atom, or is optionally terminated by a 2-pyridinyl group (e.g. to form a 2-pyridinylmethyl group);

R<sup>2</sup> represents linear C<sub>2-3</sub> alkyl;

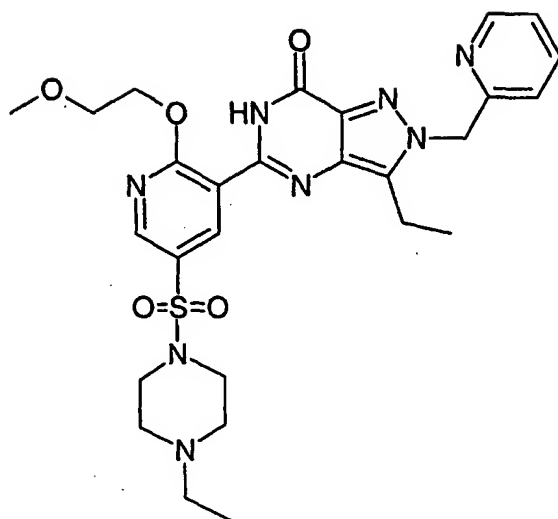
R<sup>3</sup> represents linear or branched C<sub>2-4</sub> alkyl, which alkyl group is optionally interrupted by an oxygen atom;

R<sup>4</sup> is CO<sub>2</sub>R<sup>7</sup>; or R<sup>4</sup> is a morpholinylsulphonyl, or piperazin-1-ylsulphonyl group having a substituent, R<sup>10</sup> at the 4-position of the piperazinyl group wherein R<sup>10</sup> represents methyl or ethyl.

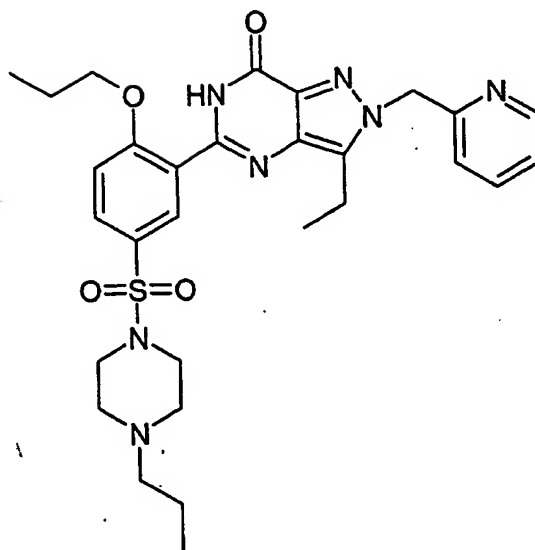
[0039] Particularly preferred compounds that may be formed according to a process of the invention include sildenafil (1A), and the following five compounds:



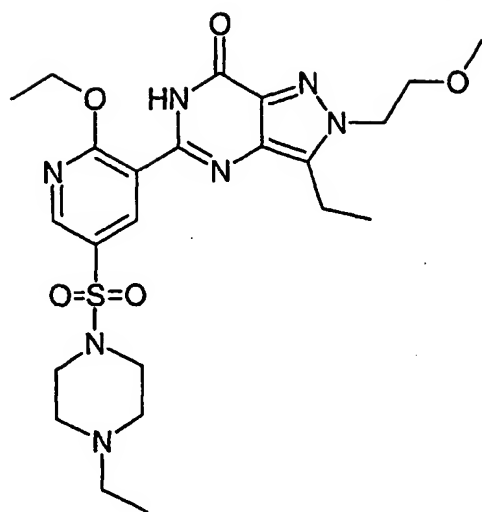
1B



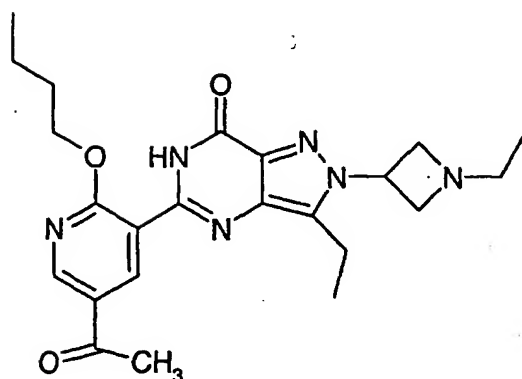
1C



1D



1E



1F

**[0040]** Said compounds 1B, 1C, 1D, 1E and 1F are otherwise known as: 1B, (+)-3-ethyl-5-[5-(4-ethylpiperazin-1-ylsulphonyl)-2-(2-methoxy-1(R)-methylethoxy)pyridin-3-yl]-2-methyl-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one also known as 3-Ethyl-5-{5-[4-ethylpiperazin-1-ylsulphonyl]-2-[(1 R)-2-methoxy-1-methylethyl]oxy}pyridin-3-yl)-2-methyl-2,6-dihydro-7H-pyrazolo[4,3-d] pyrimidin-7-one, the compound of Example 118 of WO99/54333; 1C, 3-ethyl-5-[5-(4-ethylpiperazin-1-ylsulphonyl)-2-n-propoxyphenyl]-2-(pyridin-2-yl)methyl-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one, the compound of Example 5 of WO98/49166; 1D, 3-ethyl-5-[5-(4-ethylpiperazin-1-ylsulphonyl)-2-(2-methoxyethoxy)pyridin-3-yl]-2-(pyridin-2-yl)methyl-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one, the compound of Example 4 of WO99/54333; 1E, 5-[2-Ethoxy-5-(4-ethylpiperazin-1-ylsulphonyl)pyridin-3-yl]-3-ethyl-2-[2-methoxyethyl]-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one, also known as 1-{6-ethoxy-5-[3-ethyl-6,7-dihydro-2-(2-methoxyethyl)-7-oxo-2H-pyrazolo[4,3-d]pyrimidin-5-yl]-3-pyridyl sulphonyl}-4-ethylpiperazine, the compound of Example 103 of WO 01/27113, exemplified hereinafter as Example 1; 1F, 5-[5-(5-acetyl-2-butoxy-3-pyridinyl)-3-ethyl-2-(1-ethyl-2-(1-ethyl-3-azetidinyl)-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one, the compound of Example 132 of WO 01/27112, exemplified hereinafter as Example 2.

**[0041]** Compounds may be isolated from reaction mixtures using known techniques.

**[0042]** Substituents on the aryl (e.g. phenyl), and (if appropriate) heterocyclic, group(s) in compounds defined herein

may be converted to other substituents using techniques well known to those skilled in the art. For example, amino may be converted to amido, amido may be hydrolysed to amino, hydroxy may be converted to alkoxy, alkoxy may be hydrolysed to hydroxy etc.

[0043] It will be appreciated by those skilled in the art that, in the processes described above, the functional groups of intermediate compounds may be, or may need to be, protected by protecting groups.

[0044] Functional groups which it is desirable to protect thus include hydroxy, amino and carboxylic acid. Suitable protecting groups for hydroxy include trialkylsilyl and diarylalkylsilyl groups (e.g. *tert*-butyldimethylsilyl, *tert*-butyldiphenylsilyl or trimethylsilyl), tetrahydropyranyl, benzyl and alkylcarbonyl groups (e.g. methyl- and ethylcarbonyl groups). Suitable protecting groups for amino include benzyl, *tert*-butoxycarbonyl, 9-fluorenylmethoxycarbonyl or benzyloxycarbonyl. Suitable protecting groups for carboxylic acid include C<sub>1-6</sub> alkyl, allyl or benzyl esters.

[0045] The protection and deprotection of functional groups may take place before or after any of the reaction steps described hereinbefore.

[0046] Protecting groups may be removed in accordance with techniques which are well known to those skilled in the art and as described hereinafter.

[0047] The use of protecting groups is fully described in "Protective Groups in Organic Chemistry", edited by JWF McOmie, Plenum Press (1973), and "Protective Groups in Organic Synthesis", 3<sup>rd</sup> edition, TW Greene & PGM Wutz, Wiley-Interscience (1999).

[0048] The process of the invention possesses the advantage that the intermediate pyrazoles of general formula (II) which are used in the synthesis of pyrimidin-7-ones, and in particular in the preparation of sildenafil, compound IA herein, may be prepared from commercially-available starting materials in fewer steps than in processes described in the prior art, without concomitant losses in terms of yield of key intermediates and of final compounds. Further, said pyrazoles are obtained in desirable levels of purity according to the process of the present invention.

[0049] Further, the process of the invention may have the advantage that pyrazole compounds of general formula (II) may be prepared in less time, more conveniently, and at a lower cost, than when prepared in processes described in the prior art.

[0050] The invention is illustrated, but in no way limited, by the following examples.

[0051] All <sup>1</sup>H NMR spectra were recorded using a Varian Unity 300 MHz machine.

#### Preparation 1

##### 2,2-dimethoxybutane

[0052] Methyl ethyl ketone (672 mL) was charged to a 2L round bottomed flask and stirred at room temperature before being treated with, trimethylorthoformate (763 mL) and para-toluenesulphonic acid (6.65 g, 0.5 mol%). Over a 15 min period the internal temperature rose to 46°C, so the reaction was cooled to 0°C for 30 min. The reaction was then stirred at room temperature for 2 h. The reaction was then neutralised by pouring onto sodium carbonate (ca. 750 g) with constant stirring. The resultant slurry was filtered under vacuum and the resultant filtrate was distilled at atmospheric pressure. The fraction boiling in the range 118°C-124°C was collected as a colourless liquid, 582 g, 70%.

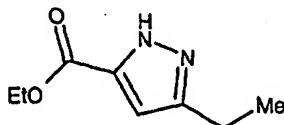
<sup>1</sup>H NMR (CDCl<sub>3</sub>): δ = 0.88 (3H, t), 1.24 (3H, s), 1.61 (2H, q), 3.17 (6H, s).

#### Example 1

N-[3-Carbamoyl-5-ethyl-1-(2-methoxyethyl)-1H-pyrazol-4-yl]-2-ethoxy-5-(4-ethyl-1-piperazinyl sulfonyl) nicotinamide.

(a) Ethyl 3-ethyl-1H-pyrazole-5-carboxylate (IIA) from (III) and (V)

[0053]



To a stirred solution of 2,2-dimethoxybutane (10 g, 84.7 mMol) in CH<sub>2</sub>Cl<sub>2</sub> (50 mL) under a nitrogen atmosphere at 0°C was added pyridine (13.7 mL, 169.5 mMol). The reaction mixture was maintained at 0°C and a solution of trichloroacetyl



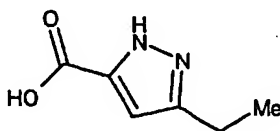
chloride (18.9 mL, 169.5 mMol) in  $\text{CH}_2\text{Cl}_2$  (35 mL) was added over 1 hour with constant stirring. The yellow-orange solution begins to precipitate a white solid as the reaction progresses. The reaction mixture is allowed to warm to room temperature over 20 h. The reaction mixture was diluted with ethanol (150 mL) and re-cooled to  $0^\circ\text{C}$  before treatment with hydrazine hydrate (8.2 mL, 169.5 mMol) as a solution in ethanol (35 mL) over 30 min. The reaction was heated to  $50^\circ\text{C}$  and solvent was distilled at atmospheric pressure. The temperature was increased until the head temperature reached  $78^\circ\text{C}$ . Reflux was maintained for a further 2 h, before cooling to room temperature. The reaction mixture was diluted with water (250 mL) and ethanol was removed by evaporation at reduced pressure. The resultant mixture was extracted with  $\text{CH}_2\text{Cl}_2$  (3 x 200 mL). The combined organics were dried ( $\text{MgSO}_4$ ), filtered and evaporated at reduced pressure to afford the title compound as a brown oil, 12.05 g, 85%.

$^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$ =1.20 (3H, t), 1.28 (3H, t), 2.67 (2H, q), 4.29 (2H, q), 6.55 (1H, s), 12.56 (1H, s).

[0054] LRMS  $m/z$ = 167.1  $[\text{M}-\text{H}]^+$ ,  $\text{C}_8\text{H}_{12}\text{N}_2\text{O}_2$  requires 168.2.

(b) Ethyl 3-ethyl-1H-pyrazole-5-carboxylic acid (IIA) from (IIA) via route 1

[0055]



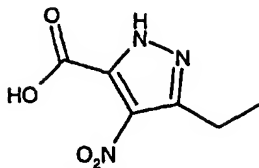
[0056] Aqueous sodium hydroxide solution (10M; 100 ml, 1.0 mol) was added dropwise to a stirred suspension of the title compound of Example (a) (66.0 g, 0.39 mol) in methanol and the resulting solution heated under reflux for 4 hours. The cool reaction mixture was concentrated under reduced pressure to ca. 200 ml, diluted with water (200 ml) and this mixture washed with toluene (3 x 100 ml). The resulting aqueous phase was acidified with concentrated hydrochloric acid to pH 4 and the white precipitate collected and dried by suction to provide the title compound (34.1 g).  $\delta$  ( $\text{DMSO}-d_6$ ): 1.13 (3H,t), 2.56 (2H,q), 6.42 (1H,s).

(c) 4-Nitro-3-n-propyl-1H-pyrazole-5-carboxylic acid

[0057] Fuming sulphuric acid (17.8 ml) was added dropwise to stirred, ice-cooled fuming nitric acid (16.0 ml), the resulting solution heated to  $50^\circ\text{C}$ , then 3-n-propyl-1H-pyrazole-5-carboxylic acid (Chem. Pharm. Bull., 1984, 32, 1568; 16.4 g, 0.106 mol) added portionwise over 30 minutes whilst maintaining the reaction temperature below  $60^\circ\text{C}$ . The resulting solution was heated for 18 hours at  $60^\circ\text{C}$ , allowed to cool, then poured onto ice. The white precipitate was collected, washed with water and dried by suction to yield the title compound (15.4 g), m.p.  $170-172^\circ\text{C}$ . Found: C, 42.35; H, 4.56; N, 21.07.  $\text{C}_7\text{H}_9\text{N}_3\text{O}_4$  requires C, 42.21; H, 4.55; N, 21.10%.  $\delta$  ( $\text{DMSO}-d_6$ ): 0.90 (3H,t), 1.64 (2H,m), 2.83 (2H,m), 14.00 (1H,s).

(d) 3-Ethyl-4-nitro-1H-pyrazole-5-carboxylic acid (IIA) to (AA) via route 2

[0058]



[0059] Obtained from the title compound of Example (b), by analogy with the process of Example (c), as a brown solid (64%).  $\delta$  ( $\text{DMSO}-d_6$ ): 1.18 (3H,t), 2.84 (2H,m), 13.72 (1H,s).

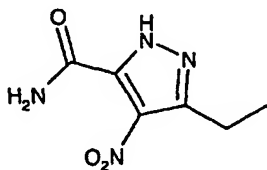
(e) 4-Nitro-3-n-propyl-1H-pyrazole-5-carboxamide

[0060] A solution of the title compound of Example (c) (15.4 g, 0.077 mol) in thionyl chloride (75 ml) was heated under reflux for 3 hours and then the cool reaction mixture evaporated under reduced pressure. The residue was azeotroped with tetrahydrofuran (2 x 50 ml) and subsequently suspended in tetrahydrofuran (50 ml), then the stirred suspension ice-cooled and treated with gaseous ammonia for 1 hour. Water (50 ml) was added and the resulting mixture evaporated under reduced pressure to give a solid which, after trituration with water and drying by suction, furnished the title compound

(14.3 g), m.p. 197-199°C. Found: C, 42.35; H, 5.07; N, 28.38.  $C_7H_{10}N_4O_3$  requires C, 42.42; H, 5.09; N, 28.27%.  $\delta$  (DMSO- $d_6$ ): 0.90 (3H,t), 1.68 (2H,m), 2.86 (2H,t), 7.68 (1H,s), 8.00 (1H,s).

(f) 3-Ethyl-4-nitro-1H-pyrazole-5-carboxamide BA from AA via route 3

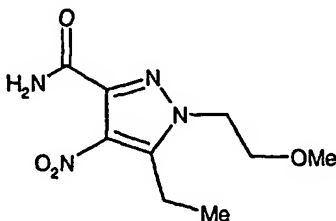
[0061]



[0062] Obtained from the title compound of Example (d), by analogy with Example (e), as a white solid (90%).  $\delta$  (DMSO- $d_6$ ): 1.17 (3H,t), 2.87 (2H,m), 7.40 (1H,s), 7.60 (1H,s), 7.90 (1H,s). LRMS:  $m/z$  185 (M+1)<sup>+</sup>.

(g)(i) 5-Ethyl-1-(2-methoxyethyl)-4-nitro-1H-pyrazole-3-carboxamide CA from BA via route 4

[0063]



[0064] A mixture of 3-ethyl-4-nitro-1H-pyrazole-5-carboxamide (2.5 kg, 13.6 Mol), sodium carbonate (1.8 Kg, 17.0 Mol) and 2-bromoethyl methyl ether (1.98 kg, 14.2 Mol) in THF (22.5 L) and water (2.5 L) was heated under reflux and stirred for 20 hours. The mixture was cooled to ambient temperature and  $CH_2Cl_2$  (67.5 L) and water (22.5 L) were added. The resultant organic and aqueous layers were separated. The aqueous phase was extracted with  $CH_2Cl_2$  (22.5 L) and the combined organic solution was distilled under atmospheric pressure and replaced with ethyl acetate (33 L) to a final volume of 17 L. The cooled mixture was granulated at ambient temperature for 2 hours, filtered and washed with ethyl acetate (2.5 L). This afforded 5-ethyl-1-(2-methoxyethyl)-4-nitro-1H-pyrazole-3-carboxamide as a white crystalline solid, 2.10 kg, 57%. m.p. = 140°C. Found: C, 44.46; H, 5.79; N, 23.01.  $C_9H_{14}N_4O_4$  requires C, 44.63; H, 5.79; N, 23.14%.

$\delta$ ( $CDCl_3$ ): 1.18 (3H, t), 2.98 (2H, q), 3.22 (3H, s), 3.77 (2H, t), 4.28 (2H, q), 6.03 (1H, s), 7.36 (1H, s). LRMS:  $m/z$  = 243 (M+1)<sup>+</sup>

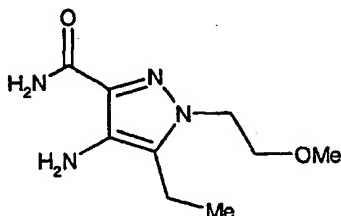
(g)(ii) 5-Ethyl-1-(2-methoxyethyl)-4-nitro-1H-pyrazole-3-carboxamide.

[0065] A mixture of 3-ethyl-4-nitro-1H-pyrazole-5-carboxamide (25 g, 0.136 Mol), sodium carbonate (18 g, 0.17 Mol) and sodium iodide (20.4 g, 0.136 Mol) were suspended in ethyl methyl ketone (125 mL) at room temperature. 2-bromoethyl methyl ether (12.8 mL, 0.142 Mol) was added and the mixture was heated to reflux and stirred for 70 hours. The mixture was cooled to ambient temperature and water (250 mL) was added. The resultant slurry was warmed to

reflux and held at that temperature for 30 min before cooling to room temperature. The resultant precipitate was granulated at room temperature for 3 h, filtered and vacuum dried to afford 5-ethyl-1-(2-methoxyethyl)-4-nitro-1H-pyrazole-3-carboxamide as a yellow crystalline solid 24.3 g, 74%. Data as reported for Example (g)(i).

(h) 4-Amino-5-ethyl-1-(2-methoxyethyl)-1H-pyrazole-3-carboxamide (IA) from CA via route 5

[0066]



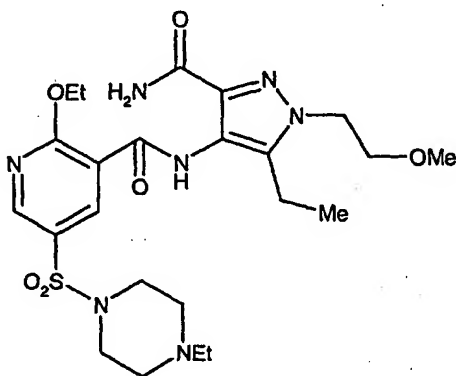
[0067] A mixture of 5-ethyl-1-(2-methoxyethyl)-4-nitro-1H-pyrazole-3-carboxamide (20 g, 82.6 mMol) and 5%Pd/C (1 g) in methanol (200 mL) was pressurised at 50psi/25°C in a sealed vessel and stirred for 15 hours. At the end of the reaction the mixture was filtered through arbolcel and the filter cake was washed with methanol. The methanolic solution was distilled at atmospheric pressure and replaced with ethyl acetate to a final volume of 100 mL. The cooled mixture was granulated at ambient temperature for 2 h filtered and washed with ethyl acetate (20 mL) to afford 4-amino-5-ethyl-1-(2-methoxyethyl)-1H-pyrazole-3-carboxamide as a white crystalline solid, 15 g, 88%. m.p. = 131°C. Found: C, 50.75; H, 7.62; N, 26.38. C<sub>9</sub>H<sub>16</sub>N<sub>4</sub>O<sub>2</sub> requires C, 50.94; H, 7.55; N, 26.42%.

$\delta$ (CDCl<sub>3</sub>): 1.20 (3H, t), 2.63 (2H, q), 3.32 (3H, s), 3.74 (2H, t), 3.95 (2H, s), 4.15 (2H, t), 5.27 (1H, s), 6.59 (1H, s).

[0068] LRMS: m/z = 213 (M+1)<sup>+</sup>

(i) N-[3-Carbamoyl-5-ethyl-1-(2-methoxyethyl)-1H-pyrazol-4-yl]-2-ethoxy-5-(4-ethyl-1-piperazinyl sulfonyl)nicotinamide.

[0069]

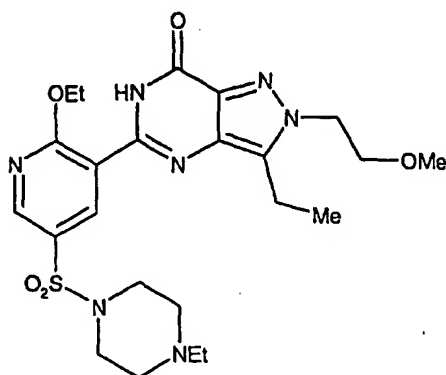


[0070] 2-ethoxy-5-(4-ethyl-1-piperazinylsulfonyl)nicotinic acid (2.31 kg, 6.73 Mol) was suspended in ethyl acetate (16.2 L) and 1,1-carbonyldimidazole (1.09 kg, 6.73 Mol) was added at room temperature. The reaction mixture was heated at 45°C for 40 minutes and then the reaction was stirred for a further 40 minutes at reflux. After cooling to ambient temperature 4-amino-5-ethyl-1-(2-methoxyethyl)-1H-pyrazole-3-carboxamide (1.5 kg, 7.06 Mol) was added to the cooled mixture, and the reaction stirred for a further 15 hours under reflux. The mixture was cooled filtered and the filter cake was washed with 90% water / 10% ethyl acetate, (2 mL / g) to afford N-[3-carbamoyl-5-ethyl-1-(2-methoxyethyl)-1H-pyrazol-4-yl]-2-ethoxy-5-(4-ethyl-1-piperazinyl sulfonyl) nicotinamide as an off white crystalline solid, 3.16 kg, 88%. m.p. = 156°C. Found: C, 51.33; H, 6.56; N, 18.36. C<sub>23</sub>H<sub>35</sub>N<sub>7</sub>O<sub>6</sub>S requires C, 51.40; H, 6.53; N, 18.25%.

$\delta(\text{CDCl}_3)$ : 1.04 (3H, t), 1.22 (3H, t), 1.60 (3H, t), 2.44 (2H, q), 2.54 (4H, m), 2.96 (2H, q), 3.12 (4H, m), 3.36 (3H, s), 3.81 (2H, t), 4.27 (2H, t), 4.80 (2H, q), 5.35 (1H, s), 6.68 (1H, s), 8.66 (1H, d), 8.86 (1H, d), 10.51 (1H, s).  
LRMS:  $m/z = 539 (\text{M}+1)^+$

(i) 1-(6-Ethoxy-5-[3-ethyl]-6,7-dihydro-2-(2-methoxyethyl)-7-oxo-2H-pyrazole[4,3-d]pyrimidin-5-yl)-3-pyridylsulfonyl)-4-ethylpiperazine•ethyl acetate solvate.

[0071]



[0072] A mixture of *N*-[3-carbamoyl-5-ethyl-1-(2-methoxyethyl)-1H-pyrazol-4-yl]-2-ethoxy-5-(4-ethyl-1-piperazinyl sulfonyl) nicotinamide (1.18 kg, 2.2 Mol), potassium *tert*-butoxide (500 g, 4.4 moles) and ethyl acetate (193 g) in ethanol (11.8 L) was heated at 120°C for 20 hours. The reaction mixture was then concentrated under reduced pressure, in total approx. 10 L of solvent were distilled. To the residue water (2.9 L) was added and the mixture stirred at room temperature while aqueous HCl was added until pH 7.5 was obtained. Ethyl acetate (7.5 L) was added and the two phase mixture was warmed to 55°C. The organic phase was separated and the aqueous phase was extracted with further ethyl acetate (3.0 L). The combined organic phases were distilled at atmospheric pressure to a final volume of 4L. The precipitated solids were granulated at 5°C for 1 h, filtered and washed with ethyl acetate (1.2 L) and dried under vacuum. This afforded 1-(6-Ethoxy-5-[3-ethyl]-6,7-dihydro-2-(2-methoxyethyl)-7-oxo-2H-pyrazole[4,3-d]pyrimidin-5-yl)-3-pyridylsulfonyl)-4-ethylpiperazine as a light yellow crystalline solid, 877 g, 78%. m.p. = 157°C. Found: C, 52.65; H, 6.46; N, 17.76.  $\text{C}_{23}\text{H}_{33}\text{N}_7\text{O}_5\text{S} \cdot 0.2 \text{ C}_2\text{H}_5\text{CO}_2\text{CH}_3$  requires C, 53.21; H, 6.49; N, 18.25%.  
 $\delta(\text{CDCl}_3)$ : 1.07 (3H, t), 1.42 (3H, t), 1.61 (3H, t), 2.44 (2H, q), 2.57 (4H, m), 3.08 (2H, q), 3.15 (4H, m), 3.32 (3H, s), 3.92 (2H, q), 4.48 (2H, q), 4.77 (2H, q), 8.65 (1H, d), 9.06 (1H, d). The spectrum also has signals that correspond to a solvate with ethyl acetate.  
LRMS:  $m/z = 520 (\text{M}+1)^+$

## Example 2

5-(5-acetyl-2-butoxy-3-pyridinyl)-3-ethyl-2-(1-ethyl-3-azetidiny)-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one.

[0073] The title compound from Preparation 2(a) (120 mg, 0.28 mmol) and cesium carbonate (274 mg, 0.84 mmol) were dissolved in *n*-butanol (4 ml), and heated at 90°C under nitrogen with molecular sieves for 96h. The mixture was then partitioned between water (10 ml) and dichloromethane (10 ml). The organic layer was separated, and the aqueous layer extracted further with dichloromethane (3 x 15 ml). The combined organic layers were dried ( $\text{MgSO}_4$ ), and concentrated *in vacuo*. The crude product was purified by flash column chromatography (95:5:0.5-90:10:1 ethyl acetate: methanol:0.88  $\text{NH}_3$  as eluents), to yield the title compound as a colourless glass (77 mg, 0.18 mmol).

m.p. 91.6-93.7°C

<sup>1</sup>H NMR (400MHz,  $\text{CDCl}_3$ ):  $\delta = 1.00$ -1.05 (m, 6H), 1.38 (t, 3H), 1.50-1.62 (m, 2H), 1.90-2.00 (m, 2H), 2.63 (s, 3H), 2.63-2.70 (m, 2H), 3.02 (q, 2H), 3.75 (t, 2H), 3.90 (t, 2H), 4.68 (t, 2H), 5.10-5.20 (m, 1H), 8.84 (s, 1H), 9.23 (s, 1H), 10.63 (br s, 1H).

LRMS (TSP - positive ion) 439 ( $\text{MH}^+$ )

Anal. Found C, 60.73; H, 7.06; N, 18.03 Calcd for  $\text{C}_{23}\text{H}_{30}\text{O}_3\text{N}_6 \cdot 0.2 \text{ MeOH} \cdot 0.1 \text{ DIPE}$ : C, 60.88; H, 7.26; N, 17.90

## Preparation of starting materials for Example 2

2(a) 5-(5-Acetyl-2-propoxy-3-pyridinyl)-3-ethyl-2-(1-ethyl-3-azetidiny)-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one

[0074] Sodium cyanoborohydride (92 mg, 1.47 mmol) was added to a stirring solution of title compound from Preparation 2(b) (500 mg, 0.98 mmol), acetaldehyde (64  $\mu$ l, 1.18 mmol) and sodium acetate (161 mg, 1.96 mmol) in methanol (10 ml) under nitrogen at room temperature. After 1h the mixture was poured into NaHCO<sub>3</sub> (sat. aq., 20 ml), and extracted with dichloromethane (3 x 15 ml). The combined organic layers were dried (MgSO<sub>4</sub>) and concentrated *in vacuo*. The crude product was purified by flash column chromatography (95:5:0.5-80:20:1 ethyl acetate:methanol:0.88 NH<sub>3</sub> as eluent) to yield the title compound as a white solid (140 mg, 0.33 mmol).

<sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>):  $\delta$  = 0.97 (t, 3H), 1.03 (t, 3H), 1.30 (t, 3H), 2.82-2.97 (m, 2H), 2.58-2.65 (m, 5H), 2.98 (q, 2H), 3.68 (t, 2H), 3.85 (dd, 2H), 4.58 (dd, 2H), 5.05-5.17 (m, 1H), 8.79 (s, 1H), 9.18 (s, 1H), 10.62 (br s, 1H).

LRMS (TSP - positive ion) 426 (MH<sup>+</sup>)

2(b) 5-(5-Acetyl-2-propoxy-3-pyridinyl)-2-(3-azetidiny)-3-ethyl-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one

[0075] The title compound of Preparation 2(c) (1.44 g, 3.0 mmol) in acetone (50 ml) and sulphuric acid (1N, 3 ml) was treated with mercuric sulphate (268 mg, 9.0 mmol) and heated to reflux for 6h. The reaction mixture was concentrated to ~20 ml *in vacuo*, poured into sodium bicarbonate (sat. aq., 20ml) and extracted into methylene chloride (6 x 20 ml). Combined organics were washed with brine (20 ml), dried over MgSO<sub>4</sub>, and concentrated to a brown oil which was taken up in 40% trifluoroacetic acid in methylene chloride (50ml) and water (1 ml) and stirred for 1h at room temperature. After evaporation *in vacuo*, the residue was purified by column chromatography (eluting with 95:5:1 methylene chloride:methanol:0.88 ammonia) to afford the title compound as a white hygroscopic foam (1.65 g).

m.p. 128.5-130.0°C

<sup>1</sup>H NMR (400MHz, MeOD):  $\delta$  = 1.00 (t, 3H), 1.30 (t, 3H), 1.79-1.90 (m, 2H), 2.60 (s, 3H), 3.00-3.10 (q, 2H), 4.50 (t, 2H), 4.60-4.70 (m, 4H), 5.65-5.78 (m, 1H), 8.65 (s, 1H), 8.90 (s, 1H)

LRMS (TSP - positive ion) 397 (MH<sup>+</sup>)

2(c) tert-Butyl 3-[3-ethyl-5-(5-ethynyl-2-propoxy-3-pyridinyl)-7-oxo-6,7-dihydro-2H-pyrazolo[4,3-d]pyrimidin-2-yl]-1-azetidinecarboxylate

[0076] Prepared from the title compound of Preparation 2(d) by the method of Preparation 2(c)(i).

<sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>):  $\delta$  = 1.05 (t, 3H), 1.30 (t, 3H), 1.43 (s, 9H), 1.88-2.00 (m, 2H), 3.00 (q, 2H), 3.19 (s, 1H), 4.35 (app t, 2H), 4.52 (app t, 2H), 4.60-4.80 (br s, 2H), 5.22 (t, 1H), 8.39 (s, 1H), 8.80 (s, 1H), 10.75 (br s, 1H)

LRMS (TSP - positive ion) 496 (MNH<sub>4</sub><sup>+</sup>).

2(c)(i) 5-(2-Butoxy-5-ethynyl-3-pyridinyl)-3-ethyl-2-(2-methoxyethyl)-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one

[0077] Potassium fluoride (22 mg, 0.38 mmol) was added to a stirred solution of the title compound of Preparation 2(d)(i) (90 mg, 0.19 mmol) in aqueous *N,N*-dimethylformamide (2 mL *N,N*-dimethylformamide /0.2 mL water) at 0°C. After 10 min the reaction was allowed to warm to room temperature and stirred for 2 h. The reaction mixture was diluted with ethyl acetate and washed with water, 1 N hydrochloric acid (3 times) and brine. The organic layer was dried (MgSO<sub>4</sub>) and concentrated to give the title compound as a white solid (75 mg).

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.00 (t, 3H), 1.40 (t, 3H), 1.50 (m, 2H), 1.90 (m, 2H), 3.05 (q, 2H), 3.20 (s, 1H), 3.30 (s, 3H), 3.85 (t, 2H), 4.40 (t, 2H), 4.60 (t, 2H), 8.40 (s, 1H), 8.80 (s, 1H), 10.70 (s, 1H).

LRMS (TSP): 396.3 (MH<sup>+</sup>).

2(d) tert-Butyl 3-(3-ethyl-7-oxo-5-(2-propoxy-5-((trimethylsilyl)ethynyl)-3-pyridinyl)-6,7-dihydro-2H-pyrazolo[4,3-d]pyrimidin-2-yl)-1-azetidinecarboxylate

[0078] Prepared from the title compound of Preparation 2(e) by the method of Preparation 2(d)(i).

<sup>1</sup>H NMR (400MHz, MeOD):  $\delta$  = 0.25 (s, 9H), 1.05 (t, 3H), 1.31 (t, 3H), 1.44 (s, 9H), 1.87-1.96 (m, 2H), 3.00 (q, 2H), 4.33 (t, 2H), 4.52 (t, 2H), 4.54-4.80 (m, 2H), 5.18-5.25 (m, 1H), 8.32 (d, 1H), 8.74 (d, 1H)

LRMS (TSP - positive ion) 569 (MNH<sub>4</sub><sup>+</sup>), 552.0 (MH<sup>+</sup>)

Anal. Found C, 60.82; H, 6.90; N, 15.15 Calcd for C<sub>28</sub>H<sub>38</sub>O<sub>4</sub>N<sub>6</sub>Si: C, 61.07; H, 6.95; N, 15.26.

2(d)(i) 5-(2-Butoxy-5-trimethylsilyl-ethynyl-3-pyridinyl)-3-ethyl-2-(2-methoxyethyl)-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one

[0079] The title compound from Example 1 of PCT application IB00/1430 (127 mg, 0.25 mmol) was suspended in triethylamine (2 mL) and trimethylsilylacetylene (38 mg, 0.39 mmol) and acetonitrile (2 mL). Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (5 mg, 0.006 mmol) and cuprous iodide (1.2 mg, 0.006 mmol) were added and the reaction mixture stirred. After 1 h a further portion of trimethylsilylacetylene (19 mg, 0.19 mmol) was added and stirring continued for 2 h. The solvent was evaporated and the residue partitioned between ethyl acetate and water. The organics were washed with brine, dried (MgSO<sub>4</sub>) and concentrated to give a brown foam. Purification by flash column chromatography (gradient elution from 100% dichloromethane to 99% dichloromethane/methanol) gave the title compound as a light brown solid (108 mg).

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ = 0.25 (s, 9H), 1.00 (t, 3H), 1.40 (t, 3H), 1.50 (m, 2H), 1.90 (m, 2H), 3.10 (q, 2H), 3.30 (s, 3H), 3.90 (t, 2H), 4.40 (t, 2H), 4.60 (t, 2H), 8.40 (s, 1H), 8.80 (s, 1H), 10.70 (s, 1H).

LRMS (TSP): 468.3 (MH<sup>+</sup>).

2(e) tert-Butyl 3-[3-ethyl-5-(5-iodo-2-propoxy-3-pyridinyl)-7-oxo-6,7-dihydro-2H-pyrazolo[4,3-d]pyrimidin-2-yl]-1-azetidinecarboxylate

[0080] The title compound was prepared from the product of Preparation 2(f) using the method of Preparation 2(e)(i). <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>): δ = 1.05 (t, 3H), 1.30 (t, 3H), 1.43 (s, 9H), 1.87-1.96 (m, 2H), 3.00 (q, 2H), 4.34 (t, 2H), 4.49 (t, 2H), 4.60 (br s, 2H), 5.20 (t, 1H), 8.41 (d, 1H), 8.94 (s, 1H), 10.75 (br s, 1H)

LRMS (TSP - positive ion) 598.1 (MNH<sub>4</sub><sup>+</sup>)

Anal. Found C, 47.54; H, 5.02; N, 14.09 Calcd for C<sub>23</sub>H<sub>29</sub>O<sub>4</sub>N<sub>6</sub>I: C, 47.60; H, 5.04; N, 14.48.

2(e)(i) 3-Ethyl-5-(5-iodo-2-propoxy-3-pyridinyl)-1-[2-(4-morpholinyl)ethyl]-1,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one

[0081] The title compound of Preparation 48 of PCT application IB00/1430 (15.78 g, 28.4 mmol) was dissolved in n-propanol (200 ml), ethyl acetate (6 ml) and potassium t-butoxide (3.2 g, 28.4 mmol) were added and the resultant mixture heated to reflux for 6h. Additional potassium t-butoxide (1.6 g, 14.2 mmol) was added and the mixture heated for a further 2h, after which the solvent was removed *in vacuo*. The residue was partitioned between water (50 ml) and methylene chloride (100 ml) and the organic phase separated. The aqueous phase was extracted with dichloromethane (2 x 100 ml) and the combined organics dried over MgSO<sub>4</sub> and reduced to a yellow solid (~17 g). Purification by column chromatography (elution with ethyl acetate) gave the title compound (13.3 g, 24.1 mmol) together with recovered starting material (2.31 g, 4.2 mmol).

m.p. 175-177°C.

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ = 1.1 (t, 3H), 1.4 (t, 3H), 1.9-2.05 (m, 2H), 2.45-2.55 (m, 4H), 2.85 (t, 2H), 3.0 (q, 2H), 3.6-3.65 (m, 4H), 4.5 (t, 2H), 4.7 (t, 2H), 8.4 (s, 1H), 9.0 (s, 1H), 10.95 (br s, 1H).

LRMS (TSP) 540 (MH<sup>+</sup>).

Analysis: found C, 46.79; H, 5.01; N, 15.44. Calcd for C<sub>21</sub>H<sub>27</sub>N<sub>6</sub>O<sub>3</sub>I: C, 46.85; H, 5.05; N, 15.61%

2(f) tert-Butyl 3-(3-(aminocarbonyl)-5-ethyl-4-[(5-iodo-2-propoxy-3-pyridinyl)carbonyl]amino)-1H-pyrazol-1-yl)-1-azetidinecarboxylate

[0082] The title compound was prepared by the method of Preparation 2(f)(i) using the products from Preparations 2(g) and 2(i).

<sup>1</sup>H NMR (400MHz, DMSO): δ = 0.95 (t, 3H), 1.05 (t, 3H), 1.40 (s, 9H), 1.78-1.88 (m, 2H), 2.68 (q, 2H), 4.22-4.35 (m, 4H), 4.40 (t, 2H), 5.33 (t, 1H), 7.35 (bs, 1H), 7.52 (bs, 1H), 8.40 (s, 1H), 8.55 (s, 1H), 10.10 (s, 1H)

LRMS (TSP - positive ion) 373.2 (MH<sup>+</sup>- BOC and I)

Anal. Found C, 45.11; H, 5.07; N, 13.56 Calcd for C<sub>23</sub>H<sub>31</sub>O<sub>5</sub>N<sub>6</sub>I. 0.2 DCM: C, 45.28; H, 5.14; N, 13.66.

2(f)(i) N-(3-(Aminocarbonyl)-1-[2-dimethylamino)ethyl]-5-ethyl-1H-pyrazol-4-yl)-2-butoxy-5-iodonicotinamide

[0083] Cesium carbonate (1.17 g, 3.59 mmol) was added to a stirred solution of the title compound from Preparation 16 of PCT application IB00/1430 (800 mg, 1.79 mmol) and *N,N*-dimethylaminoethyl chloride hydrochloride (309 mg, 2.15 mmol) in *N,N*-dimethylformamide (10 mL) under a nitrogen atmosphere. The mixture was heated at 80°C for 24 h. The mixture was cooled and extracted from water with ethyl acetate. The organics were dried (MgSO<sub>4</sub>) and concentrated to give a brown oil. Purification by flash column chromatography (gradient elution from 100% dichloromethane to 90% dichloromethane/MeOH) gave the product as a pale brown oil (522 mg).

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ = 0.95 (t, 3H), 1.20 (t, 3H), 1.40 (m, 2H), 1.90 (m, 2H), 2.35 (s, 6H), 2.80 (t, 2H), 2.85 (q, 2H), 4.20 (t, 2H), 4.60 (t, 2H), 5.30 (br s, 1H), 6.60 (br s, 1H), 8.40 (s, 1H), 8.75 (s, 1H), 10.35 (s, 1H).  
LRMS (TSP): 529.5 (MH<sup>+</sup>).

5 2(g) N-[3-(Aminocarbonyl)-5-ethyl-1H-pyrazol-4-yl]-5-iodo-2-propoxynicotinamide

[0084] The title compound was prepared from 2-propoxy-5-iodonicotinic acid (see Preparation 2(h) and 4-amino-3-ethyl-1H-pyrazole-5-carboxamide (prepared as described in WO 98/49166) according to the method described in Preparation 2(g)(i).

10 <sup>1</sup>H NMR (300 MHz, d<sub>4</sub>-MeOH): δ = 1.0 (t, 3H), 1.25 (t, 3H), 1.85-2.0 (m, 2H), 2.8 (q, 2H), 4.5 (t, 2H), 8.5 (s, 1H), 8.6 (s, 1H).  
LRMS (TSP) 444 (MH<sup>+</sup>).

15 2(g)(i) N-[3-(Aminocarbonyl)-5-ethyl-1-(2-methoxyethyl)-1H-pyrazol-4-yl]-2

butoxy-5-iodonicotinamide

[0085] Oxalyl chloride (2 g, 15.9 mmol) was added to a stirred solution of the title compound from Preparation 4 of PCT application IB00/1430 (1.28 g, 3.98 mmol) in dichloromethane (20 mL) and 3 drops *N,N*-dimethylformamide added. After 2.5 h the solvent was evaporated and the residue azeotroped 3 times with dichloromethane. The residue was resuspended in dichloromethane (4 mL) and added to a stirred mixture of the title compound of Preparation 11 from PCT application IB00/1430 (0.76 g, 3.58 mmol) and triethylamine (0.8 g, 7.97 mmol) in dichloromethane (10 mL). After 1 h the solvent was evaporated and the residue partitioned between ethyl acetate and water. The organic phase was separated and washed with 2N HCl (twice), sodium bicarbonate solution (twice) and brine before being dried (MgSO<sub>4</sub>) and concentrated. The product was triturated with ether and filtered to give 820 mg of pure product as a white solid. The mother liquor was concentrated and purified by flash column chromatography (elution with 80% ethyl acetate : hexane), to give a further 605 mg of product.

25 <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ = 0.95 (t, 3H), 1.20 (t, 3H), 1.45 (m, 2H), 1.90 (m, 2H), 2.85 (q, 2H), 3.35 (s, 3H), 3.80 (t, 2H), 4.25 (t, 2H), 4.60 (t, 2H), 5.20 (br s, 1H), 6.60 (br s, 1H), 8.40 (s, 1H), 8.80 (s, 1H), 10.30 (s, 1H).

30 LRMS (TSP): 516.2 (MH<sup>+</sup>).

2(h) 2-Propoxy-5-iodonicotinic acid

[0086] The title compound was prepared from 2-propoxy nicotinic acid (prepared as described in WO 99/54333, the compound 2-n-propoxypyridine-3-carboxylic acid, Preparation 46 prepared by the process of Preparation 1) using the method of Preparation 2(h)(i).

35 <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ = 1.05 (t, 3H), 1.85-2.0 (m, 2H), 4.5 (t, 2H), 8.5 (s, 1H), 8.6 (s, 1H).

Analysis: found C, 35.16; H, 3.19; N, 4.46. Calcd for C<sub>9</sub>H<sub>10</sub>INO<sub>3</sub>: C, 35.19; H, 3.28; N, 4.56%

40 2(h)(i) 2-isoButoxy-5-iodo nicotinic acid

[0087] *N*-Iodosuccinamide (18.22 g, 0.08 mol), trifluoroacetic acid (100 mL) and trifluoroacetic anhydride (25 mL) were added to 2-isobutoxynicotinic acid (10.55 g, 0.054 mol). The mixture was refluxed for 2.5 h, cooled and the solvents evaporated. The residue was extracted from water with ethyl acetate and the organics washed with water (twice) and brine (twice), dried (MgSO<sub>4</sub>) and concentrated. The red residue was redissolved in ethyl acetate washed with sodium thiosulfate solution (twice), water (twice), brine (twice), redried (MgSO<sub>4</sub>) and concentrated to give the desired product as a yellow solid.

45 <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ = 1.05 (d, 6H), 2.20 (m, 1H), 4.40 (d, 2H), 8.50 (s, 1H), 8.70 (s, 1H).

50 LRMS (TSP): 322.3 (MH<sup>+</sup>).

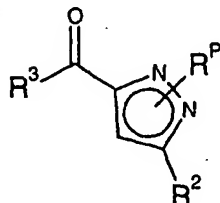
2(i) *tert*-Butyl 3-iodo-1-azetidinecarboxylate

[0088] A mixture of *tert*-butyl 3-[(methylsulfonyl)oxy]-1-azetidinecarboxylate (prepared as described in *Synlett* 1998, 379; 5.0 g, 19.9 mmol), and potassium iodide (16.5 g, 99.4 mmol) in *N,N*-dimethylformamide (25 mL), was heated at 100°C for 42 h. The cooled mixture was partitioned between water and ethyl acetate, and the layers separated. The organic phase was dried over MgSO<sub>4</sub>, concentrated under reduced pressure and the residue azeotroped with xylene. The crude product was purified by flash column chromatography (dichloromethane as eluant) to give the title compound, 3.26 g.

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ = 1.43 (s, 9H), 4.28 (m, 2H), 4.46 (m, 1H), 4.62 (m, 2H).  
LRMS (TSP) 284 (MH)<sup>+</sup>

# Claims

1. A process for the production of pyrazole compounds of general formula (II):



II

wherein R<sup>P</sup> represents H or R<sup>1</sup>;

R<sup>1</sup> represents: H, C<sub>1</sub>-C<sub>6</sub> alkyl; C<sub>1</sub>-C<sub>6</sub> alkoxy; C<sub>3</sub>-C<sub>6</sub> cycloalkyl; C<sub>1</sub>-C<sub>6</sub> alkyl(C<sub>1</sub>-C<sub>6</sub> alkoxy), Het, C<sub>1</sub>-C<sub>6</sub> alkylHet, aryl or C<sub>1</sub>-C<sub>6</sub> alkylaryl, which latter eight groups are all optionally substituted (and/or, in the case of C<sub>1</sub>-C<sub>6</sub> alkyl, optionally terminated) by one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, C(O)NR<sup>4</sup>R<sup>5</sup>, C(O)R<sup>6</sup>, C(O)OR<sup>7</sup>, OR<sup>8</sup>, NR<sup>9a</sup>R<sup>9b</sup> and SO<sub>2</sub>NR<sup>10a</sup>R<sup>10b</sup>;

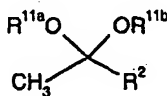
R<sup>2</sup> represents C<sub>1</sub>-C<sub>6</sub> alkyl; C<sub>1</sub>-C<sub>6</sub> alkoxy; C<sub>3</sub>-C<sub>6</sub> cycloalkyl; C<sub>1</sub>-C<sub>6</sub> alkyl(C<sub>1</sub>-C<sub>6</sub> alkoxy), Het, C<sub>1</sub>-C<sub>6</sub> alkylHet, aryl or C<sub>1</sub>-C<sub>6</sub> alkylaryl, which latter eight groups are all optionally substituted (and/or, in the case of C<sub>1</sub>-C<sub>6</sub> alkyl, optionally terminated) by one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, C(O)NR<sup>4</sup>R<sup>5</sup>, C(O)R<sup>6</sup>, C(O)OR<sup>7</sup>, OR<sup>8</sup>, NR<sup>9a</sup>R<sup>9b</sup> and SO<sub>2</sub>NR<sup>10a</sup>R<sup>10b</sup>;

R<sup>3</sup> represents: OH, C<sub>1</sub>-C<sub>6</sub> alkoxy or NR<sup>4</sup>R<sup>5</sup>;

R<sup>4</sup>, R<sup>5</sup>, R<sup>6</sup>, R<sup>7</sup>, R<sup>8</sup>, R<sup>10a</sup> and R<sup>10b</sup> independently represent H or C<sub>1</sub>-C<sub>6</sub> alkyl;

R<sup>9a</sup> and R<sup>9b</sup> either independently represent, H or C<sub>1</sub>-C<sub>6</sub> alkyl or, together with the nitrogen atom to which they are attached, represent azetidiny, pyrrolidinyl or piperidinyl,

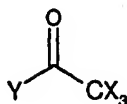
which process comprises the reaction of a compound of general formula (III),



III

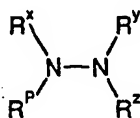
wherein R<sup>11a</sup> and R<sup>11b</sup> independently represent C<sub>1</sub>-C<sub>6</sub> alkyl, R<sup>2</sup> is as defined herein before, with an acylating agent of the formula (IV) in the presence of a base,





IV

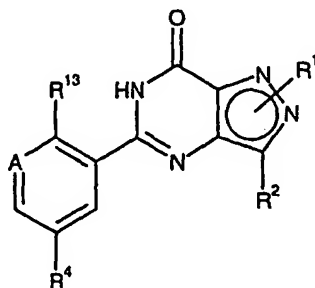
wherein X represents halogen independently selected from Cl, F or Br and wherein Y represents halogen or OR<sup>12</sup> wherein R<sup>12</sup> is C<sub>1</sub>-C<sub>6</sub> alkyl, C(O)CX<sub>3</sub>, Het, or C<sub>0</sub>-C<sub>6</sub> alkyl(Het) wherein Het is pyridine or imidazole, followed by *in situ* reaction with a hydrazine compound of general formula (V)



V

wherein R<sup>p</sup> represents H or R<sup>1</sup> wherein R<sup>1</sup> is as defined hereinbefore, and R<sup>x</sup>, R<sup>y</sup> and R<sup>z</sup> are independently selected from H or electron donating groups (EDG) or electron withdrawing groups (EWG) wherein said electron withdrawing or donating groups are labile under the conditions of the reaction, wherein said process is optionally carried out in the presence of an activating agent.

2. A process for the preparation of compounds of formula (I):



(I)

or a pharmaceutically or veterinarily acceptable salt thereof, or a pharmaceutically or veterinarily acceptable solvate of either entity, wherein:

A is CH or N;

R<sup>1</sup> represents: H, C<sub>1</sub>-C<sub>6</sub> alkyl; C<sub>1</sub>-C<sub>6</sub> alkoxy; C<sub>3</sub>-C<sub>6</sub> cycloalkyl; C<sub>1</sub>-C<sub>6</sub> alkyl(C<sub>1</sub>-C<sub>6</sub> alkoxy), Het, C<sub>1</sub>-C<sub>6</sub> alkylHet, aryl or C<sub>1</sub>-C<sub>6</sub> alkylaryl, which latter eight groups are all optionally substituted (and/or, in the case of C<sub>1</sub>-C<sub>6</sub> alkyl, optionally terminated) by one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, C(O)NR<sup>4</sup>R<sup>5</sup>, C(O)R<sup>6</sup>, C(O)OR<sup>7</sup>, OR<sup>8</sup>, NR<sup>9a</sup>R<sup>9b</sup> and SO<sub>2</sub>NR<sup>10a</sup>R<sup>10b</sup>;

R<sup>2</sup> represents C<sub>1</sub>-C<sub>6</sub> alkyl; C<sub>1</sub>-C<sub>6</sub> alkoxy; C<sub>3</sub>-C<sub>6</sub> cycloalkyl; C<sub>1</sub>-C<sub>6</sub> alkyl(C<sub>1</sub>-C<sub>6</sub> alkoxy), Het, C<sub>1</sub>-C<sub>6</sub> alkylHet,

aryl or C<sub>1</sub>-C<sub>6</sub> alkylaryl, which latter eight groups are all optionally substituted (and/or, in the case of C<sub>1</sub>-C<sub>6</sub> alkyl, optionally terminated) by one or more substituents selected from halo, cyano, nitro, C<sub>1</sub>-C<sub>6</sub> alkyl, C(O)NR<sup>4</sup>R<sup>5</sup>, C(O)R<sup>6</sup>, C(O)OR<sup>7</sup>, OR<sup>8</sup>, NR<sup>9a</sup>R<sup>9b</sup> and SO<sub>2</sub>NR<sup>10a</sup>R<sup>10b</sup>;

R<sup>1</sup> represents: NR<sup>p</sup>R<sup>q</sup>;

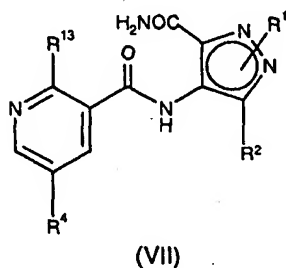
R<sup>p</sup>, R<sup>q</sup>, R<sup>6</sup>, R<sup>7</sup>, R<sup>8</sup>, R<sup>10a</sup> and R<sup>10b</sup> independently represent H or C<sub>1</sub>-C<sub>6</sub> alkyl;

R<sup>9a</sup> and R<sup>9b</sup> either independently represent, H or C<sub>1</sub>-C<sub>6</sub> alkyl or, together with the nitrogen atom to which they are attached, represent azetidiny, pyrrolidiny or piperidiny;

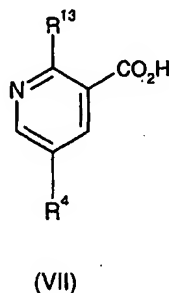
R<sup>4</sup> is C<sub>1</sub>-C<sub>4</sub> alkyl optionally substituted with OH, NR<sup>5</sup>R<sup>6</sup>, CN, CONR<sup>5</sup>R<sup>6</sup> or CO<sub>2</sub>R<sup>7</sup>; or R<sup>4</sup> is CO<sub>2</sub>R<sup>7</sup>; or R<sup>4</sup> is a pyrrolidinylsulphonyl, piperidinylsulphonyl, morpholinylsulphonyl, or piperazin-1-ylsulphonyl group having a substituent, R<sup>10</sup> at the 4-position of the piperazinyl group wherein said piperazinyl group is optionally substituted with one or two C<sub>1</sub> to C<sub>4</sub> alkyl, C<sub>1</sub> to C<sub>3</sub> alkoxy, NR<sup>7</sup>R<sup>8</sup> or CONR<sup>7</sup>R<sup>8</sup> groups and is optionally in the form of its 4-N-oxide;

R<sup>10</sup> represents H or C<sub>1</sub>-C<sub>6</sub> alkyl;

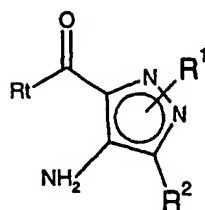
wherein said compounds may be prepared from compounds of general formula (VIII):



wherein R<sup>1</sup>, R<sup>2</sup>, R<sup>4</sup> and R<sup>13</sup> are as defined hereinbefore wherein said compound of general formula (VIII) is prepared from the reaction of a compound of general formula (VI):

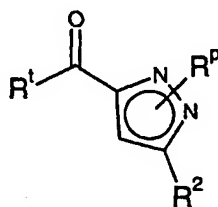


wherein R<sup>4</sup> and R<sup>13</sup> are as defined herein before via coupling with a compound of general formula (VI):



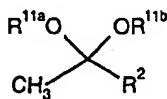
VI

wherein  $R^1$  and  $R^2$  are as defined hereinbefore and wherein  $R^t$  is  $NR^pR^q$  wherein  $R^p$  and  $R^q$  independently represent H or  $C_1-C_6$  alkyl and wherein said compound of general formula (VI) is prepared by nitration and hydrogenation of a compound of general formula (II):



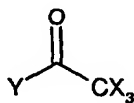
II

wherein  $R^t$  and  $R^2$  are as defined hereinbefore and wherein  $R^p$  represents  $R^1$  as defined hereinbefore **characterised in that** said compound of general formula (II) is prepared from the reaction of a compound of general formula (III),



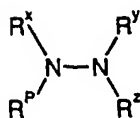
III

wherein  $R^{11a}$  and  $R^{11b}$  independently represent  $C_1-C_6$  alkyl,  $R^2$  is as defined herein before, with an acylating agent of the formula (IV) in the presence of a base,



IV

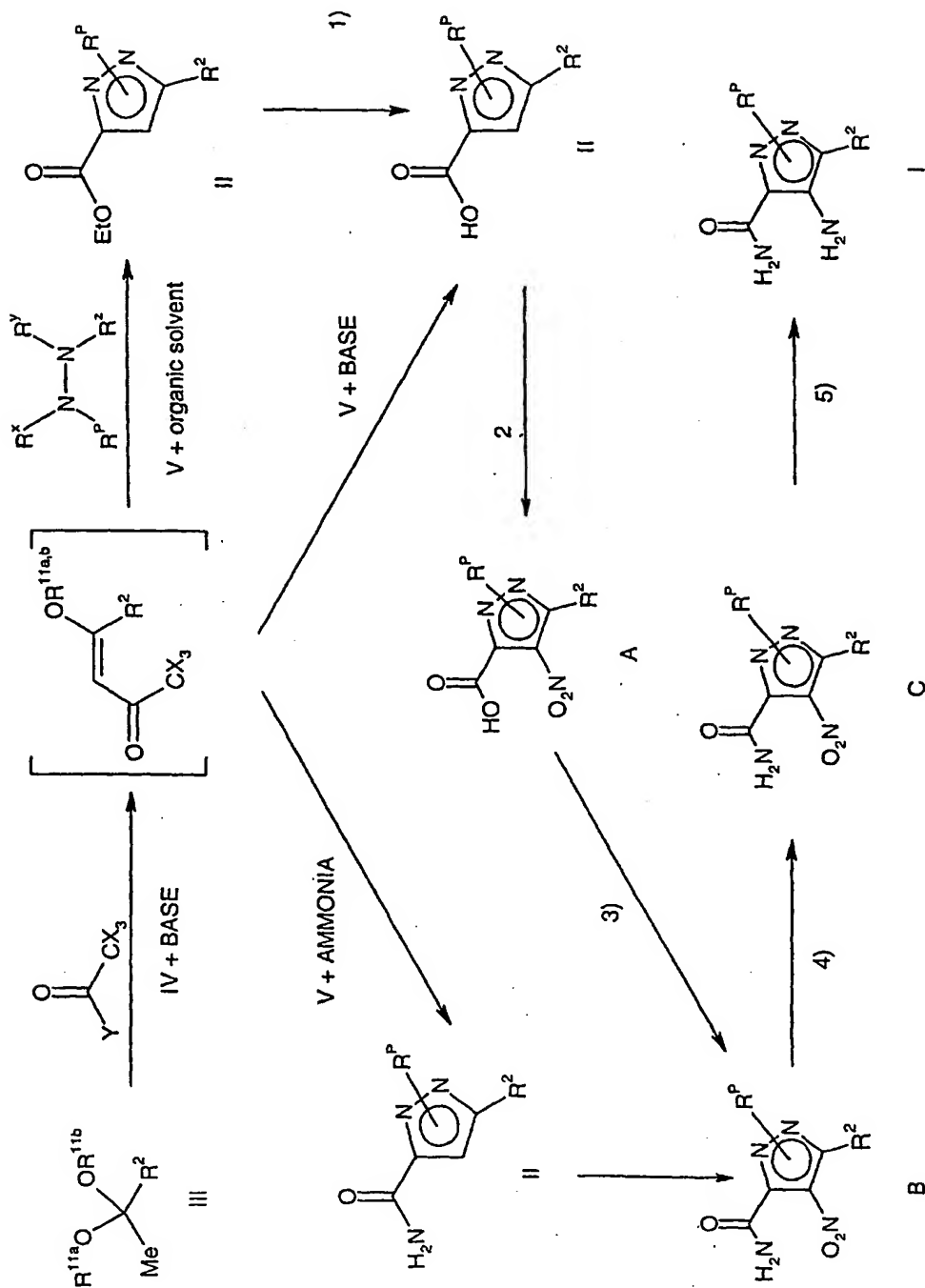
wherein X represents halogen independently selected from Cl, F or Br, and Y represents halogen or  $OR^{12}$  wherein  $R^{12}$  is:  $C_1-C_6$  alkyl;  $C(O)CX_3$ ; Het; or  $C_0-C_6$  alkyl(Het) wherein Het is pyridine or imidazole, followed by *in situ* reaction with a hydrazine compound of general formula (V)



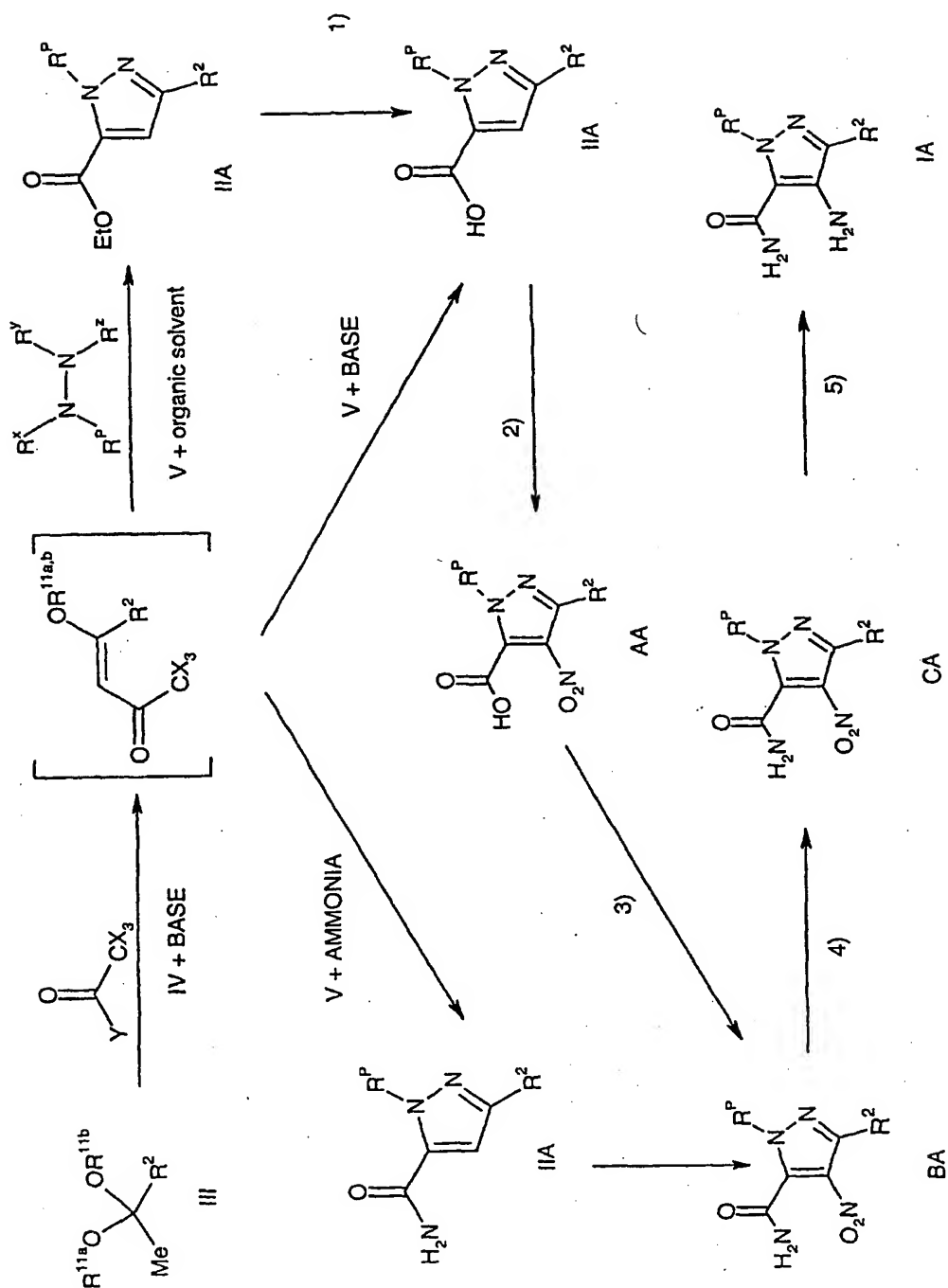
V

wherein  $R^p$  represents H or  $R^1$  wherein  $R^1$  is as defined hereinbefore, and  $R^x$ ,  $R^y$  and  $R^z$  are independently selected from H or electron donating groups (EDG) or electron withdrawing groups (EWG) wherein said electron withdrawing or donating groups are labile under the conditions of the reaction, wherein said process for the preparation of compound (II) is optionally carried out in the presence of an activating agent.

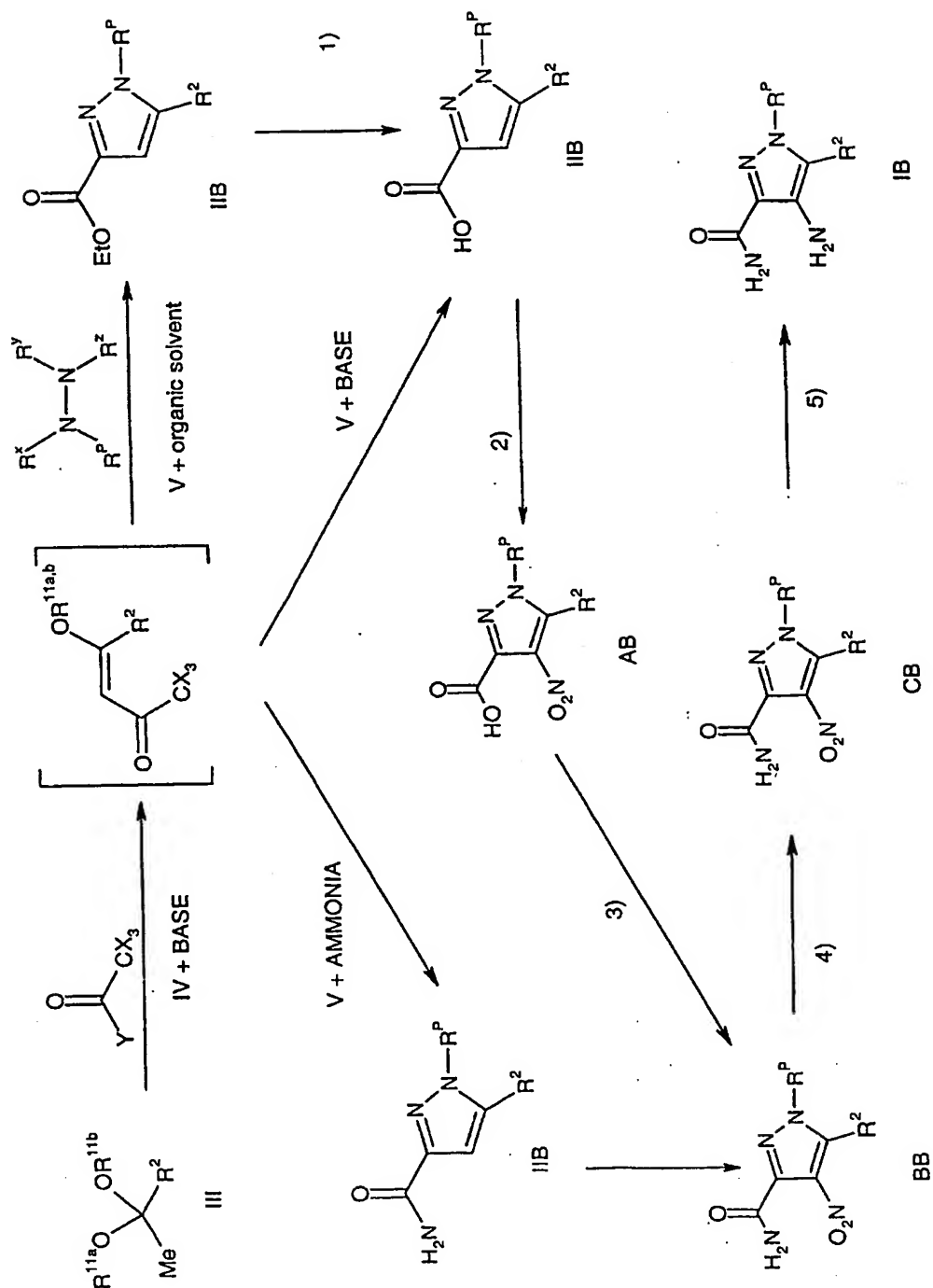
3. A process according to claim 2 wherein the compound of formula (I) is sildenafil, 5-[2-Ethoxy-5-(4-ethylpiperazin-1-ylsulphonyl)pyridin-3-yl]-3-ethyl-2-[2-methoxyethyl]-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one, or 5-(5-acetyl-2-butoxy-3-pyridinyl)-3-ethyl-2-(1-ethyl-2-(1-ethyl-3-azetidiny)-2,6-dihydro-7H-pyrazolo[4,3-d]pyrimidin-7-one.



Scheme 1



Scheme 2



Scheme 3



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## EUROPEAN SEARCH REPORT

Application Number  
EP 01 30 6346

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